Isolation, structure elucidation and antibacterial activities of streptothricin acids

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Five streptothricin acids (compounds 1–5) were isolated by ion-exchange resin chromatography and preparative RP-HPLC from the fermentation broth of *Streptomyces qinlingensis*. Their structures were elucidated mainly by analyses of the IR, HR-EIS-MS and NMR spectral data. They were deduced as two known compounds, streptothricin F acid (1) and streptothricin D acid (2), and three new compounds, 12-carbamoylstreptothricin E acid (3), 12-carbamoylstreptothricin D acid (4) and N-acetyl-streptothricin D acid (5), respectively. The antibacterial activities of 1–5 against *Escherichia coli, Bacillus subtilis, Staphylococcus aureus, Bacillus cereus* and *Pseudomonas aeruginosa* were assayed by micro-broth dilution. Comparison of the MICs with streptothricin F and D showed that the antimicrobial activities of 1–5 were decreased significantly but varied with the structures.

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Keywords: antibacterial activity; streptothricin acid; structural elucidation

Streptothricin antibiotic is a type of broad-spectrum antibiotics. As streptothricin F was isolated from Streptomyces lavendulae in 1942,¹ dozens of streptothricins (STs) have been isolated from natural resources. Although they were shown to possess strong antimicrobial activity against many species of fungi and bacteria, their application in the therapeutic area has been hampered because of the nephrotoxicity.²⁻⁵ Hamano et al.⁶ isolated a novel ST-resistance gene (sttH) and showed that the hydrolysis of the amide bond of streptolidine lactam could be catalyzed by the enzyme in vitro. With the breaking of the amide bond, the antimicrobial activity of streptothricin D (ST-D) was altered from broad-spectrum to bacteria-specific, and its toxicity against eukaryotic cells was also reduced at the same time. These results suggest that ST-D acid is a potential candidate for clinical development or use as a new lead compound for drug discovery. Attempts at preparing ST acids were made from STs chemically, but only ST-F acid, which bears one β-lysine, was readily obtained, whereas others that bear more β -lysine residues were not prepared successfully.^{7,8} In addition, only the bio-activities of ST-F and ST-D acids have been investigated.⁶

In the process of screening for new antibiotics, *S. qinlingensis*, a ST producer, was isolated from soil samples. ST-F, ST-D and two other novel compounds, 12-carbamoyl-ST-F and 12-carbamoyl-ST-D, have been isolated from its fermentation broth in our laboratory.⁹ In subsequent research, a facile LC-MS/MS method based on RP-HPLC coupled with electrospray ionization tandem mass spectrometry was used for the analysis of STs in the fermentation broth of *S. qinlingensis*.¹⁰ A total of 19 ST-like compounds were identified or tentatively characterized on the basis of their mass spectral data; among them, 12 were ST acids. Here, we report the

isolation, structure elucidation and antibacterial activities of five ST acids that were isolated by ion-exchange resin chromatography and preparative RP-HPLC from the fermentation broth of *S. qinlingensis*. These compounds were identified as two known compounds, ST-F acid (1) and ST-D acid (2), and three new compounds, 12-carbamoylstreptothricin E acid (3), 12-carbamoylstreptothricin D acid (4) and N-acetylstreptothricin D acid (5). The antibacterial activities of 1–5 against *Escherichia coli, Bacillus subtilis, Staphylococcus aureus, Bacillus cereus* and *Pseudomonas aeruginosa* were assayed by means of micro-broth dilution. The results indicated that 1–5 exhibited less antibacterial activities when compared with that of ST-D, but the extent of decrease varied with the structures.

RESULTS AND DISCUSSION

Physicochemical properties of 3-5

Compound **3**: colorless amorphous powder, m.p. 132–134 °C (dec), $[\alpha]_D^{25}$ –14.0 (c 0.1, MeOH). IR ν_{max} cm⁻¹: 3382, 1651, 1555, 1492, 1388, 1070 cm⁻¹. ¹H and ¹³C NMR: see Tables 1 and 2. High-resolution electrospray ionization mass spectometry (HR-ESI-MS) (*m/z*): 649.3630 [M+H]⁺, calcd for C₂₅H₄₉N₁₀O₁₀, 649.3628.

Compound 4: colorless amorphous powder, m.p. 150–152 °C (dec), $[\alpha]_D^{25}$ +8 (c 0.1, MeOH). IR ν_{max} cm⁻¹: 3396, 1655, 1558, 1492, 1390, 1067 cm⁻¹; ¹H and ¹³C NMR: see Tables 1 and 2. HR-ESI-MS (*m/z*): 777.4576 [M+H]⁺, calcd for C₃₁H₆₂N₁₂O₁₁, 777.4577.

Compound 5: colorless amorphous powder, m.p. 172–174 °C (dec), $[\alpha]_D^{25}$ +4.2 (c 0.1, MeOH). IR ν_{max} cm⁻¹: 3418, 1652, 1554, 1445, 1388, 1071 cm⁻¹; ¹H and ¹³C NMR: see Tables 1 and 2. HR-ESI-MS (*m/z*): 819.4683 [M+H]⁺, calcd for C₃₃H₆₂N₁₂O₁₂, 819.4683.

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Table 1	ιH	NMR	data	of	compounds	1–5	(500 MHz,	in	D_2O	
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Position	1	2	3	4	5
2	4.58 (d, <i>J</i> =5)	4.69 (d, <i>J</i> =5)	4.63 (d, <i>J</i> =5)	4.61 (d, <i>J</i> =5)	4.71 (d, <i>J</i> =5)
3	4.27 (m)	4.34 (m)	4.30 (m)	4.30 (m)	4.36 (m)
4	4.14 (m)	4.14 (m)	4.15 (m)	4.14 (m)	4.15 (m)
5	3.28 (dd, <i>J</i> =5, 14)	3.25 (dd, <i>J</i> =5, 14)	3.28 (m)	3.29 (dd, <i>J</i> =3, 14)	3.29 (m)
5	3.12 (dd, <i>J</i> =10, 14)	3.13 (dd, <i>J</i> =11, 14)	3.13 (m)	3.13 (dd, <i>J</i> =11, 14)	3.13 (m)
7	5.14 (d, <i>J</i> =10)	5.14 (d, <i>J</i> =10)	5.09 (d, <i>J</i> =10)	5.14 (d, <i>J</i> =10)	5.12 (d, <i>J</i> =10)
8	4.25 (dd, <i>J</i> =3, 10)	4.26 (dd, <i>J</i> =3, 10)	4.31 (dd, <i>J</i> =3, 10)	4.26 (dd, <i>J</i> =3, 10)	4.26 (dd, J=3, 10)
9	4.18 (t, <i>J</i> =3)	4.15 (t, <i>J</i> =3)	4.10 (t, <i>J</i> =3)	4.18 (t, <i>J</i> =3)	4.08 (t, <i>J</i> =3)
10	4.79	4.79	3.88 (d, <i>J</i> =4)	3.88 (d, <i>J</i> =4)	4.79
11	4.33 (t, <i>J</i> =6)	4.33 (t, <i>J</i> =6)	4.33 (t, <i>J</i> =6)	4.33 (t, <i>J</i> =6)	4.32 (t, <i>J</i> =6)
12	3.74 (d, <i>J</i> =6)	3.74 (d, <i>J</i> =6)	4.23 (d, <i>J</i> =5)	4.23 (d, <i>J</i> =5)	3.72 (d, <i>J</i> =6)
12	3.74 (d, <i>J</i> =6)	3.74 (d, <i>J</i> =6)	4.26 (d, <i>J</i> =7)	4.26 (d, <i>J</i> =7)	3.72 (d, <i>J</i> =6)
15	2.81 (dd, <i>J</i> =5, 17)	2.79 (dd, <i>J</i> =5, 17)	2.78 (dd, <i>J</i> =5, 17)	2.80 (dd, <i>J</i> =5, 17)	2.54 (dd, <i>J</i> =5, 17)
15	2.71 (dd, <i>J</i> =8, 17)	2.65 (dd, <i>J</i> =8, 17)	2.65 (dd, <i>J</i> =8, 17)	2.71 (dd, <i>J</i> =8, 17)	2.43 (dd, J=8, 17)
16	3.71 (m)	3.66 (m)	3.68 (m)	3.68 (m)	4.14 (m)
17	1.79 (m)	1.81 (m)	1.74 (m)	1.72 (m)	1.54 (m)
18	1.79 (m)	1.70 (m)	1.68 (m)	1.70 (m)	1.54 (m)
19	3.05 (m)	3.12 (m)	3.24 (m)	3.07 (m)	3.19 (m)
21		2.72 (dd, <i>J</i> =5, 17)	2.75 (dd, <i>J</i> =5, 17)	2.75 (dd, <i>J</i> =5, 17)	2.75 (dd, <i>J</i> =5, 17)
21		2.63 (dd, <i>J</i> =8, 17)	2.62 (dd, <i>J</i> =8, 17)	2.62 (dd, <i>J</i> =8, 17)	2.65 (dd, <i>J</i> =8, 17)
22		3.66 (m)	3.68 (m)	3.66 (m)	3.64 (m)
23		1.71 (m)	1.82 (m)	1.75 (m)	1.70 (m)
24		1.68 (m)	1.63 (m)	1.66 (m)	1.65 (m)
25		3.24 (m)	3.07 (m)	3.23 (m)	3.24 (m)
27		2.70 (dd, <i>J</i> =5, 17)		2.72 (dd, <i>J</i> =5, 17)	2.72 (dd, <i>J</i> =5, 17)
27		2.60 (dd, <i>J</i> =8, 17)		2.62 (dd, <i>J</i> =8, 17)	2.62 (dd, J=8, 17)
28		3.67 (m)		3.69 (m)	3.67 (m)
29		1.78 (m)		1.81 (m)	1.82 (m)
30		1.64 (m)		1.62 (m)	1.82 (m)
31		3.26 (m)		3.24 (m)	3.07 (m)
N-acetyl					1.99 (s)

Structure elucidation

Two known compounds (1, 2) and three novel compounds (3-5) were obtained from the fermentation broth of *S. qinlingensis* finally. Compounds 1 and 2 were identified as ST-F acid and ST-D acid on the basis of the M.P., IR, MS and NMR data.^{7,8}

Compound 3 was obtained as a colorless amorphous powder. The molecular formula C25H48N10O10 was determined by HR-ESI-MS at m/z 649.3630 [M+H]⁺ (calcd for C₂₅H₄₉N₁₀O₁₀, 649.3628). The IR spectra showed the characteristic absorption bands at 1654 and 1556 cm⁻¹, which revealed the existence of a peptide bond. The NMR spectral data (Tables 1 and 2) suggested the presence of two amido groups, one carboxylic acid group, one carbamoyl group and one guanidino group. The spectral characteristics were very similar to those of STs. By comparing the ¹H and ¹³C NMR data of **3** with those of ST-D acid (2), we found that the chemical shift of C-4, C-5 in the ¹³C NMR spectra and the chemical shift of H-4 and H-5 in the ¹H NMR spectra were not different. Compound 3 was assumed to be the ST acid type compound. The hypothesis is also proved by the results of ESI-MS/MS; m/z 171, which was assigned to streptolidine, was a characteristic ion peak of STs, whereas the corresponding ion was observed at m/z 189 in the spectrum of 3. This indicated that streptolidine lactam was converted into acid for compound 3. The differences between the ¹H and ¹³C NMR data of 3 and 2 were also observed; we found a chemical shift of C-9 downfield 2.6 p.p.m., C-10 upfield 2.2 p.p.m., C-11 upfield 1.1 p.p.m. and C-12 downfield

3.6 p.p.m. in the ¹³C NMR spectrum, as well as a chemical shift of H-10 upfield 0.9 p.p.m. and H-12 downfield 0.5 p.p.m. in the ¹H NMR spectrum. These differences may be attributed to the variety of the substitution position of the carbamoyl group attached to D-gulosamine in which the substitution position of carbamoyl was transferred from C-10 to C-12. This presumption was confirmed by the heteronuclear multiple-bond correlation (HMBC) experiment. In the HMBC spectra of 2, the protons of H-9, H-10 and H-11 were correlated with the carbonyl of the carbamoyl group. However, the carbonyl of the carbamoyl group was correlated with the protons of H-10, H-11 and H-12 in the HMBC spectra of 3. Thus, the carbamoyl group of 3 was substituted at C-12, and the compound was identified as 12-carbamoylstreptothricin E acid. To our best knowledge, only four 12-carbamoylstreptothricins, 12-carbamoylstreptothricin F, D, C and B, have been reported; so compound 3 is a novel 12-carbamoylstreptothricin antibiotic (Figure 1).9,11

Compound **4** was obtained as a colorless amorphous powder. The molecular formula $C_{31}H_{60}N_{12}O_{11}$ was determined by HR-ESI-MS at m/z 777.4576 [M+H]⁺ (calcd for $C_{31}H_{61}N_{12}O_{11}$, 777.4577). The IR spectra showed the characteristic absorption bands at 1654 and 1556 cm⁻¹, which revealed the existence of a peptide bond. The NMR spectral data (Tables 1 and 2) suggested the presence of three amido groups, one carboxylic acid group, one carbamoyl group and one guanidino group. By comparing the ¹H and ¹³C NMR data of **4** with those of **2** and **3**, we found that the chemical shift of C-4, C-5,

Table 2 ¹³C NMR data of compounds 1–5 (125 MHz, in D₂O)

Position	1	2	3	4	5
1	173.53	173.45	172.05	174.26	173.10
2	56.74	56.73	57.05	57.18	56.83
3	61.33	61.31	61.54	61.55	61.27
4	67.97	67.96	67.96	67.98	67.96
5	40.59	40.61	40.61	40.61	41.32
6	158.46	157.84	158.26	158.35	158.46
7	79.04	79.09	79.14	79.14	79.13
8	48.95	48.92	48.68	48.92	48.93
9	66.42	66.46	69.03	67.98	66.74
10	69.98	69.99	67.76	67.76	69.97
11	73.37	73.38	72.27	72.28	73.32
12	60.19	60.21	63.79	63.79	60.26
13	157.85	158.45	158.99	158.99	157.82
14	171.97	172.11	171.82	172.08	173.82
15	36.26	36.80	36.61	36.58	41.32
16	48.25	48.67	48.46	48.43	47.17
17	28.97	29.46	29.41	29.37	31.37
18	22.85	24.28	24.27	24.25	24.85
19	38.89	38.76	38.77	38.79	38.96
20		171.82	171.62	171.62	171.62
21		36.59	36.82	36.58	36.58
22		48.66	48.49	48.43	48.46
23		29.38	29.07	29.06	29.42
24		22.88	22.89	24.25	24.30
25		38.76	38.91	38.79	36.77
26		171.61		171.62	171.62
27		36.56		36.58	36.85
28		48.45		48.43	48.70
29		29.06		29.06	29.07
30		22.89		22.90	22.90
31		38.89		38.88	39.00
N-acetyl					173.10, 22.10

C-10, C-12 in the ^{13}C NMR spectra, and the chemical shift of H-4, H-5, H-10, H-12 in the ^1H NMR spectra were similar; so **4** also belongs to 12-carbamoylstreptothricin acid. The molecular weight of **4** is 128 Da higher than that of **3**, which corresponds to a molecule of β -lysine. On the basis of the ^1H and ^{13}C NMR data and HMBC spectra, **4** was identified as 12-carbamoylstreptothricin D acid.

Compound 5 was obtained as a colorless amorphous powder. The molecular formula $\mathrm{C}_{33}\mathrm{H}_{62}\mathrm{N}_{12}\mathrm{O}_{12}$ was determined by HR-ESI-MS at m/z 819.4683 [M+H]⁺ (calcd for C₃₃H₆₃N₁₂O₁₂, 819.4683). The IR spectra showed the characteristic absorption bands at 1651 and 1554 cm⁻¹, which revealed the existence of a peptide bond. The NMR spectral data (Tables 1 and 2) suggested the presence of four amido groups, one carboxylic acid group, one carbamoyl group and one guanidino group. By comparing the ¹H and ¹³C NMR data of 5 with those of 2–4, we found that the chemical shift of C-4, C-5 in the ¹³C NMR spectrum, and H-4, H-5 in the ¹H NMR spectrum were similar to 3; the chemical shift of C-10, C-12 in the ¹³C NMR spectrum, and H-10, H-12 in the ¹H NMR spectrum were similar to 2; so 5 belongs to ST acid. The ¹H NMR spectrum of 5 showed a singlet (3H) at δ 1.99 in addition to all the signals observed in that of 2, and the signal of the methyl group was observed at δ 22.10 in the ¹³C NMR spectrum of 5. In the HMBC spectrum of 5, we could find that the methyl group is attached to a carbon (δ 173.10) of carbonyl group, which was absent in the ¹³C NMR spectrum of 2. The protons of H-15 and H-16 were correlated with the carbonyl group at δ 173.10, indicating that the acetyl group was attached to the amino group of C-16. Thus, **5** was identified as N-acetylstreptothricin D acid.

Streptothricin F acid is the most investigated compound of ST acids, and was first prepared from ST-F by partial hydrolysis, whereas the other ST acids bearing more β -lysine residues were not readily obtained chemically.^{7,8} The results may be attributed to that, although STs possess easily hydrolyzable bonds, such as ester and amido bonds, but the hydrolysis is strongly influenced by the number of β -lysine. In 2006, Hamano *et al.*⁶ isolated a novel ST-resistance gene (sttH) from *S. albulus*, which could catalyze the transformation of ST-D and ST-F from lactam to acid *in vitro*. The results of HPLC-MS/MS analysis also indicated the distribution of ST acids in the metabolites of *S. qinlingensis*. We therefore hypothesized that a similar resistance gene may exist in this actinomycete, the research of which is under way in our laboratory.

Antibacterial assay

Antimicrobial activities of compounds 1-5 against B. subtilis, S. aureus, E. coli, B. cereus and P. aeruginosa were determined by the micro-broth dilution method, and the results are showed in Table 3. In a comparison of the MICs of STs F and D with those of 1-5, the antimicrobial activities of 1-5 were decreased significantly, but the extents varied with the structures. The structural activity relationship of STs has been described in previous research, but that of ST acids was not investigated thoroughly.¹²⁻¹⁵ To our best knowledge, only the antibacterial activities of ST acids F and D were reported up to now. ST-D acid exhibited higher activity against B. subtilis and S. aureus than did ST-F acid, and this indicates that the number of β-lysine plays an important role in the antibacterial activity for this type of ST acid antibiotics. The activity of 4 was slightly lower than that of 2, and the only difference between them is the substituted position of the carbamoyl group in the D-gulosamine moiety, which implies that their activities were also influenced by the substituted position of the carbamoyl group. It is noticeable that acetylation of β-lysine almost resulted in the loss of activity, and this phenomenon was identical with that of classical ST antibiotics.12,13

METHODS

Microorganism and fermentation

The producing strain *S. qinlingensis* was isolated from a soil sample collected in Qinling Mountain, Shannxi Province, China, and identified by its morphology, physiology, biochemistry and 16S rRNA gene sequence. The voucher specimen of this streptomycete was deposited at the China General Microbiological Culture Collection Center as CGMCC1381, and the 16S rDNA sequence was registered in GenBank with the accession no: AM167521 in the National Center for Biological Information.

The spores of *S. qinlingensis* grown on Gause's No. 1 agar were used to inoculate into a 250 ml flask containing 50 ml of a sterile seed medium consisting of glucose 0.8%, soluble starch 0.8%, beef extract 0.6%, peptone 1.0%, and NaCl 0.5%, pH 7.2. The flask was shaken on a shaker at 210 r.p.m. for 24 h at 28 °C. Ten milliliters of the seed culture were transferred to 250 ml flasks containing 50 ml of a sterile producing medium consisting of glucose 3.0%, millet steep liquor 1.0%, peptone 1.5%, NaCl 0.5%, and CaCO₃ 0.5%, pH 7.2. Fermentation was carried out at 210 r.p.m. for 96 h at 28 °C on a rotary shaker.

Isolation and purification

Ten liters of harvest fermentation broth were adjusted to pH 3.5 with oxalic acid and stirred for 30 min. The acidic broth was filtered, adjusted to pH 8.0, and passed through a column of HD-2 (Na⁺, 2 l; Shanghai Huazhen Sci. & Tch. Co., Shanghai, China). Active principle adsorbed on the column was eluted



Table 3 Antibacterial in vitro activities of 1–5 by micro-broth dilution

_	MIC ($\mu g \cdot m l^{-1}$)						
Compounds	B. subtilis (1.0088)	S. aureus (1.0089)	E. coli (1.1636)	B. cereus (1.1846)	P. aeruginosa (1.2031)		
Streptothricin F	6.3	12.5	3.1	25.0	50.0		
Streptothricin D	3.1	6.3	3.1	25.0	50.0		
1	25.0	50.0	50.0	>100.0	>100.0		
2	12.5	25.0	12.5	25.0	50.0		
3	25.0	>100.0	25.0	50.0	>100.0		
4	25.0	50	12.5	50.0	>100.0		
5	50.0	>100.0	50.0	>100.0	>100.0		

with 0.5 $_{\rm M}$ HCl (81). The eluent was adjusted to pH 7.0 with NaHCO₃ and concentrated *in vacuo* to a small volume (about 300 ml). The concentrated solution was diluted with MeOH (3:l) to precipitate sodium chloride. The MeOH was removed on a rotary evaporator, and the aqueous solution was applied on a column of CM-Sephadex C-25 (0.51, Na⁺). The column was eluted with a linear gradient of water and 1.0 $_{\rm M}$ NaCl solution (11 each). The eluents were collected in 50 ml fractions and analyzed by HPLC-MS/MS. The analyses were performed on a Finnigan LCQ Advantage MAX LC/MS (Thermo Finnigan, San Jose, CA, USA) equipped with a Waters (Waters, Milford, MA, USA) Atlantis dC18 column (4.6 \times 250 mm, 5 μ m), and a water/TFA/acetonitrile gradient was used as mobile phase. The gradient started with 5% acetonitrile in 0.1% aqueous TFA for 10 min. The percent of acetonitrile was

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increased to 20% in the next 10 min. The flow rate of the mobile phase was 0.8 ml min⁻¹ and the injection volume of sample was 10 µl. The eluate from the HPLC was introduced into a splitter to provide a 0.1 ml min⁻¹ flow into the mass spectrometer. An electrospray ionization (ESI) interface with positive-ion mode was used. The ESI conditions were as follows: source voltage 4.5 kV, capillary temperature 300 °C, sheath gas flow 75, auxiliary gas flow 10, capillary voltage 10V. A data-dependent acquisition was used in the LC-MS/MS experiments. The collision energy for MS/MS was adjusted to 35% and the isolation width of precursor ions was 3.0 mass units. The results showed that ST acids were distributed in fractions 25-28 and 72-77, and these fractions were combined with fractions A and B, respectively. The two fractions were further desalted by chromatography on active carbon columns and concentrated under reduced pressure to a small volume (about 10 ml). The concentrated fractions were further separated on a Shimadzu (Tokyo, Japan) 6AD HPLC apparatus equipped with a column of Hypersil ODS-BP (20×250 mm, 10 µm, flow rate 3.5 ml min⁻¹), monitored by a Shodex (Kawasaki, Japan) RI-101 detector. Fraction A was eluted with 6.0% MeOH containing 2.0% TFA; fraction B was eluted with 8.0% MeOH containing 6.0% TFA. Five fractions were collected according to the peaks in the HPLC profiles, and each fraction was lyophilized, redissolved in water and passed through a column packed with 717 anion ion exchange resin (Cl-, 50 ml; Shanghai Huazhen Sci. & Tch. Co.) to exchange TFA for chloride. The solutions were lyophilized to yield five compounds, with 1 (45 mg), 2 (83 mg) and 3 (51 mg) being from fraction A, and 4 (32 mg) and 5 (36 mg) being from fraction B, respectively.

Structure elucidation

The structures of these compounds were elucidated on the basis of extensive 1D and 2D NMR experiments and high-resolution mass spectrometry, and confirmed by comparison of their ESI-MS/MS data with those of LC-MS/MS. Melting points were measured on an X4 apparatus and uncorrected. IR spectra

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were determined on an IR-450 instrument (KBr plate). ¹H, ¹³C NMR, DEPT, COSY, heteronuclear single quantum coherence and HMBC spectra were taken on a Bruker Avance 500 MHz (500 MHz for ¹H and 125 MHz for ¹³C, respectively) spectrometer (Bruker Biospin GmbH, Rheinstetten, Germany) in D₂O solution with TMS as an internal standard. The ESI-MS/MS and HR-ESI-MS spectra were obtained on a Finnigan LCQ LC-MSⁿ (Thermo Finnigan) and a Bruker APEX II mass spectrometer (Bruker Biospin GmbH) using glycerol as the matrix. Optical rotation was measured in methanol solution on a Perkin-Elmer 341 Polarimeter (Perkin Elmer, Fremont, CA, USA).

Antibacterial assay

Minimum inhibitory concentration of compounds 1-5 against B. subtilis, S. aureus, E. coli, B. cereus and P. aeruginosa were tested by the micro-broth dilution method.¹⁶ The inoculum was prepared by suspending several colonies from an overnight culture of tested bacteria from 5% sheep blood agar medium in Mueller-Hinton broth, and adjusting to a 0.5 McFarland standard (approximately $1.5{\times}10^8$ cells ml^-1). A further dilution of 1:200 was made by placing 0.25 ml of the adjusted suspension into 49.75 ml of Mueller-Hinton broth. Stock solutions of tested compounds and streptomycin (positive control) in sterile water were prepared at a concentration of $1000\,\mu g\,ml^{-1}$ and used immediately or stored in working samples at $-20\,^\circ C$ until used. Doubling dilutions of the tested compounds were prepared in Mueller-Hinton broth. All antimicrobial solutions were prepared in large volumes (50 ml); 0.1 ml samples of the antibiotic solutions and 0.1 ml inoculated suspension of the test bacterium were delivered to wells of a 96-well plate. The final concentration of inoculum in each well was 3.7×10^5 cells ml⁻¹. Minimum inhibitory concentration end points were read after 18 h of incubation at 35 °C, and were defined as the lowest concentration of antibiotics that resulted in no bacterial growth as indicated by the ODs at 650 nm. The blank control tube contained only bacteria. Three replications were carried out for each sample.

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Pentanol derivatives from basidiomycete *Catathelasma imperiale* and their 11β-hydroxysteroid dehydrogenases inhibitory activity

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Five new secondary metabolites derived from pentanol, namely catathelasmols A–E (1–5), were isolated from the fruiting bodies of the basidiomycete *Catathelasma imperiale*. Their structures were elucidated on the basis of spectroscopic analysis, and the absolute configurations were determined by computational chemistry. Compounds 3, 4 and 5 showed inhibitory activities against two isozymes of 11 β -hydroxysteroid dehydrogenases (11 β -HSD1 and 11 β -HSD2), with IC₅₀ values of 28.7–62.3 µg ml⁻¹ (human 11 β -HSD1), 30.4–149.2 µg ml⁻¹ (mouse 11 β -HSD1), 5.1–177 µg ml⁻¹ (human 11 β -HSD2) and 32.3–129.1 µg ml⁻¹ (mouse 11 β -HSD2), which catalyze the interconversion of cortisol and cortisone.

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INTRODUCTION

Catathelasma imperiale (Fr.) Sing. (Tricholomataceae) is a coniferloving basidiomycete defined by its large size, white spore print, sticky brownish cap, mealy odor and double ring. This mushroom is mainly distributed in the southwest of China. Only a few ergosterols have been reported from the chemical investigation of this fungus.¹ As a part of our efforts to discover the structurally diverse and biologically active secondary metabolites from higher fungi,²⁻⁵ the investigation of the fruiting bodies of C. imperiale has led to the isolation of five new compounds, catathelasmols A-E (1-5). Herein, details of the isolation and structural elucidation of 1-5 are described, including assignment of the absolute configurations by computational chemistry. Among them, compounds 3-5 showed inhibitory activities against two isozymes of 11B-hydroxysteroid dehydrogenases (11B-HSD1 and 11β-HSD2), which catalyze the interconversion of cortisol and cortisone. Although these secondary metabolites are structurally simple, there are very few reports on these types of natural products.

MATERIALS AND METHODS

General

Optical rotations were obtained on a Horiba SEPA-300 polarimeter (Horiba, Tokyo, Japan). IR spectra were taken on a Bruker Tensor 27 FT-IR spectrometer (Bruker GmbH, Ettlingen, Germany) with KBr pellets. NMR spectra were recorded with a Bruker DRX-500 instrument (Bruker GmbH) in $CDCl_3$

 $(\delta_{\rm H}=7.26 \text{ p.p.m.}, \delta_{\rm C}=77.00 \text{ p.p.m.})$ at room temperature. EI-MS, electrospray ionization mass spectrum (ESI-MS) and high resolution electrospray ionization mass spectrum (HR-ESI-MS) were measured on Finnigan-MAT 90 (Finnigan, Somerset, NJ, USA) and API QSTAR Pulsar i (MDS Sciex, Concord, ON, Canada) mass spectrometers, respectively. Silica gel (200–300 mesh; Qingdao Marine Chemical Inc., Qingdao, China) and Sephadex LH-20 (Amersham Biosciences, Uppsala, Sweden) were used for column chromatography. Fractions were monitored by TLC and spots were visualized by heating silica gel plates sprayed with vanillin-H₂SO₄ in ethanol.

Fungus material

The fresh fruiting bodies of *C. imperiale* were purchased at a market in Nanhua County of Yunnan Province, China, in August 2005 and were identified by Professor Mu Zang. The voucher specimen (HFG 05112) was deposited in the herbarium of Kunming Institute of Botany, Chinese Academy of Sciences.

Extraction and isolation

The dry fruiting bodies of *C. imperiale* (700 g) were extracted thrice with EtOAc (total 91) at room temperature for 3 days each time. The extract was filtered and concentrated under reduced pressure to give a residue (18.8 g), which was subjected to silica gel column chromatography eluted with CHCl₃/MeOH (from 100:0 to 0:100). The fraction (6.6 g) eluted with pure CHCl₃ was subjected to further silica gel column chromatography using a gradient of petroleum ether:acetone (150:1; 50:1; 20:1) followed by pure MeOH to give subfractions A–D. Subfraction A (300 mg) (petroleum ether:acetone 150:1, v/v) was further isolated over a silica gel column eluted with petroleum ether:

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acetone (150:1) to give a residue (160 mg) mainly containing 1 and 2, which was repeatedly subjected to silica gel column chromatography to yield 1 (40.0 mg) and 2 (1.7 mg). Subfraction B (180 mg) (petroleum ether:acetone 50:1, v/v), containing mainly 3 and 4, was further isolated on a silica gel column eluted with CHCl₃ to yield 3 (14.6 mg) and 4 (74.1 mg). Subfraction D (121 mg) eluted with MeOH was purified by Sephadex LH-20 (CHCl₃/MeOH 1:1, v/v) and silica gel column chromatography using CHCl₃/MeOH (150:1, v/v) as the eluent, to yield 5 (61.4 mg).

Computational methods

The stable geometries of 1 with low energy were investigated using HyperChem 7.0 (HyperCube, Gainesville, FL, USA). These low-energy conformations were then optimized at the B3LYP/6-31G (d) level again. The B3LYP/6-31G (d)optimized structures were then used for optical rotation calculations at the B3LYP/aug-cc-pVDZ//6-31+G (d). The calculated optical rotation for (R) configuration was +14.5°. This value is very close to the experimental magnitude of +10.6°. Thus, the absolute configuration of (+)-1 was assigned as (R). Compound 5 has a linear structure and it is difficult to use the above method to compute its optical rotation to determine the absolute configuration. Our recent matrix method was used in the study. The value of the determinant (det(D)) for (R)-5 was -2.50. According to the principle of matrix prediction, the det(D) and k_0 value were needed to use both. As the optical rotation was -10.4 (c 0.40, CHCl₃), the calculated k_0 value for this chiral secondary alcohol was 4.2, which is located in the window of coefficients of chiral secondary alcohols in chloroform. Thus, compound 5 was assigned (R) configuration.

Physicochemical properties

Catathelasmol A (1): amorphous powder; $[\alpha]_D^{26}$ +10.6 (*c* 0.30, CHCl₃); IR (KBr): 3450, 1175, 1115, 1058, 1033 cm⁻¹; ¹H- and ¹³C-NMR: see Table 1; ESI-MS (pos.): 101 [M–H₂O+H]⁺, 83 [M–2H₂O+H]⁺; HR-ESI-MS (pos.): 101.0600 ([M–H₂O+H]⁺, calcd. 101.0602).

Catathelasmol B (2): colorless oil; $[\alpha]_{D}^{26}$ +8.9 (*c* 0.30, CHCl₃); IR (KBr): 3441, 1191, 1167, 1104, 1062, 1034 cm⁻¹; ¹H- and ¹³C-NMR: see Table 1; ESI-MS (pos.): 218 [M]⁺, 201 [M-H₂O+H]⁺, 183 [M-2H₂O+H]⁺; HR-ESI-MS (pos.): 201.1125 ([M-H₂O+H]⁺, calcd. 201.1126).

Catathelasmol C (3): colorless oil; IR (KBr): 1736, 1661, 1371, 1236, 1191 cm⁻¹. 1 H- and 13 C-NMR: see Table 2; ESI-MS (pos.): 225 [M+Na]⁺; HR-ESI-MS (pos.): 225.0739 ([M+Na]⁺, calcd. 225.0738).

Catathelasmol D (4): colorless oil; IR (KBr): 3441, 2966, 1738, 1369, 1248, 1045 cm⁻¹; ¹H- and ¹³C-NMR: see Table 2; EI-MS: 160 (M⁺, 2), 129 (42), 112 (17), 100 (14), 87 (100); HR-FAB-MS (pos.): 161.0838 ([M+H]⁺, calcd. 161.0814).

Catathelasmol E (5): colorless oil; $[\alpha]_D^{26}$ –10.4 (*c* 0.40, CHCl₃); IR (KBr): 3458, 1740, 1450, 1247 cm⁻¹. ¹H- and ¹³C-NMR: see Table 2; ESI-MS (pos.): 227 [M+Na]⁺; HR-ESI-MS (pos.): 227.0889 ([M+Na]⁺, calcd. 227.0895).

Biological testing

The inhibitory activities of the compounds on human or mouse 11 β -HSD1 and 11 β -HSD2 enzymatic activities were determined by the scintillation proximity assay using microsomes containing 11 β -HSD1 or 11 β -HSD2, according to our earlier studies.⁶ Briefly, the full-length cDNAs of human or murine 11 β -HSD1 and 11 β -HSD2 were isolated from the cDNA libraries provided by the NIH Mammalian Gene Collection and cloned into pcDNA3

expression vector. HEK-293 cells were transfected with the pcDNA3-derived expression plasmid and selected by cultivation in the presence of 700 μg ml⁻¹ of G418. The microsomal fraction overexpressing 11β-HSD1 or 11β-HSD2 was prepared from the HEK-293 cells stably transfected with either 11β-HSD1 or 11β-HSD2 and was used as the enzyme source for scintillation proximity assay. Microsomes containing human or mouse 11β-HSD1 were incubated with NADPH and [³H]cortisone. Then the product, [³H]cortisol, was specifically captured by a monoclonal antibody coupled to protein A-coated scintillation proximity assay beads. The 11β-HSD2 screening was performed by incubating 11β-HSD2 microsomes with [³H]cortisol and NAD⁺ and monitoring substrate disappearance. IC₅₀ values were calculated by using Prism Version 4 (GraphPad Software, San Diego, CA, USA).

RESULTS AND DISCUSSION

Catathelasmol A (1), obtained as an amorphous powder, has a molecular formula of C5H10O3 based on the positive-ion HR-ESI-MS, showing a quasi-molecular ion peak at m/z 101. 0600 (calcd. for $[C_5H_{10}O_3-H_2O+H]^+$, 101.0602) and requiring only one degree of unsaturation. The IR spectrum showed the presence of one or more hydroxyl groups (3450 cm⁻¹). The ¹³C-NMR spectrum (Table 1) exhibited five signals: one quaternary carbon bearing two oxygens at δ 102.7, two oxymethylenes at δ 68.0 (t), 65.2 (t) and two up-field methylenes at δ 33.8 (t), 23.5 (t). It was obvious that only one degree of unsaturation was attributed to a ring. The ¹H-NMR spectrum (Table 1) showed eight protons: two oxygenated methylenes at δ 3.45 (1H, d, J=11.5 Hz), 4.14 (1H, d, J=11.5 Hz), 3.97 (2H, t, J=6.9 Hz), and up-field resonances at δ 1.61 (1H, m), 1.94 (1H, m), 1.88 (1H, m), 2.05 (1H, m). The above NMR data suggested that 1 possessed a tetrahydrofuran moiety connected with a hydroxyl and a hydroxymethyl group. The heteronuclear multi-bond correlations (HMBC) (Figure 1) from H-6 to C-2 and C-3, and from H-5 to C-2 and C-3 were observed; consequently, the hydroxyl and hydroxymethyl groups were doubtless both emplaced at C-2. The absolute configuration of (+)-1 was assigned as (R) using the B3LYP/aug-cc-pVDZ//6-31+G(d) methods7-9 based on the comparison of experimental optical rotation (+10.6°) and calculated optical rotation $(+14.5^{\circ})$. Therefore, the structure of 1 was determined as (R)-(+)-2-(hydroxymethyl)-tetrahydrofuran-2-ol and named catathelasmol A, as shown in Figure 2.

Catathelasmol B (2), a colorless oil, was obtained as a minor constituent with a molecular formula of $C_{10}H_{18}O_5$, based on the positive-ion HR-ESI-MS: 201.1125 (calcd. for $[C_{10}H_{18}O_5-H_2O+H]^+$, 201.1126). In the NMR spectra (Table 1) of 2, signals for the number of protons and carbon that were observed were only half of the number that would correspond to the molecular formula. This indicated that 2 is a symmetrical structure. The NMR data were considerably in accordance with those of 1, but their TLC behavior was discriminable, which suggested that 2 was unambiguously a dimer of 1, and there were just two possible condensed positions: C-2 or C-6 hydroxyl. There was no reaction and no corresponding product obtained in acetylation, which suggested that there was no free

Table 1	NMR spectra	data for d	compounds 🛾	1 and 2	2 in CDCl ₃
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No		1	2		
140.	$\delta_{\mathcal{C}}$	δ_H	$\delta_{\mathcal{C}}$	δ_H	
2	102.7 (s)		103.8 (s)		
3	33.8 (t)	1.61, 1.94 (each 1H, m)	33.7 (t)	1.77, 2.11 (each 1H, m)	
4	23.5 (t)	1.88, 2.05 (each 1H, m)	24.1 (t)	1.89, 2.03 (each 1H, m)	
5	68.0 (t)	3.97 (2H, 6.9)	67.9 (t)	3.94, 4.00 (each 1H, m)	
6	65.2 (t)	3.45 (1H, d, 11.5), 4.14 (1H, d, 11.5)	66.3 (t)	3.55 (1H, d, 11.8), 3.94 (1H, d, 11.8)	

No.		3		4		5		
	δ_C	δ _Η	δ_C	δ _H	δ_C	δ _H		
1	67.9 (t)	4.65 (2H, s)	63.1 (t)	4.26 (2H, s)	68.6 (t)	3.95 (1H, dd, 11.3, 7.2), 4.11 (1H, dd, 11.3, 7.2)		
2	202.9 (s)		208.9 (s)		69.3 (d)	3.84 (1H, m)		
3	35.1 (t)	2.50 (2H, t, 7.2)	34.4 (t)	2.52 (2H, t, 7.2)	29.6 (t)	1.71, 1.80 (each 1H, m)		
4	22.2 (t)	1.95 (2H, m)	22.3 (t)	1.97 (2H, m)	24.6 (t)	1.52 (2H, m)		
5	66.3 (t)	4.06 (2H, t, 6.3)	67.8 (t)	4.08 (2H, t, 6.3)	64.2 (t)	4.08 (2H, t, 6.5)		
a_1	170.2 (s)				171.3 (s)			
a ₂	20.4 (q)	2.16 (3H, s)			20.9 (q)	2.08 (3H, s)		
b_1	171.0 (s)		170.9 (s)		171.3 (s)			
b ₂	20.9 (q)	2.04 (3H, s)	20.5 (q)	2.04 (3H, s)	21.0 (q)	2.03 (3H, s)		

Table 2 NMR spectral data for compounds 3–5 in CDCl₃



Figure 1 Key HMBC correlations of compounds 1, 3 and 5.



Figure 2 Structures of compounds 1-5.

primary hydroxyl group in the structure of **2**, but that **2** was formed from two molecules of **1** through the intermolecular dehydration at C-6 hydroxyl. Thus, the structure of **2** was proposed as (2R,2'R)-2,2'-oxybis(methylene)bis(tetrahydrofuran-2-ol), as shown in Figure 2.

Catathelasmol C (3) was isolated as a colorless oil possessing the molecular formula C9H14O5, based on the positive-ion HR-ESI-MS: 225.0739 (calcd. for C₉H₁₄O₅Na, 225.0738). The IR spectrum showed the absorption bands of ester carbonyl (1736 cm¹) and keto carbonyl (1661 cm⁻¹) groups. The ¹³C-NMR spectrum (Table 2) exhibited nine carbon resonances, including one keto carbonyl at δ 202.9 (s), two oxymethylenes at δ 67.9 (t), 66.3 (t), two up-field methylene carbons at δ 35.1 (t), 22.2 (t), as well as characteristic signals at δ 170.2 (s), 20.4 (q); 171.0 (s), 20.9 (q) contributed to two acetoxyls. The ¹H-NMR spectrum (Table 2) of 3 showed signals for two oxygenated methylenes at δ 4.65 (2H, s), 4.06 (2H, t, J=6.3 Hz), two up-field methylenes at δ 2.50 (2H, t, J=7.2 Hz), 1.95 (2H, m) together with two acetoxyl methyl singlets at δ 2.16 (3H, s), 2.04 (3H, s). The above NMR character allowed us to conclude that 3 was a diacetylated pentanediol containing a keto group. By analysis of the HMBC spectrum (Figure 1), the position of the ketone was determined at C-2, in which the correlations from H-1 to C-3 and C-a1, from H-5 to C-3 and C-b1 and

from H-4 to C-2 were observed. Therefore, the structure of **3** was elucidated as 2-oxopentane-1,5-diyl diacetate named catathelasmol C, which was a new natural product.¹⁰

Catathelasmol D (4), also obtained as a colorless oil, has a molecular formula of $C_7H_{12}O_4$, based on the EI-MS showing a molecular ion peak at m/z 160, in combination with the ¹³C-NMR (DEPT) spectrum. The NMR data (Table 2) of 4 were similar to those of 3, but there was only one set of acetoxyl signals: δ 170.9 (s), 20.5 (q). Considering that the signal at δ 4.26 (2H, s, H-1) in 4 evidently shifted up-field (δ =0.39 p.p.m.) compared with that of 3, the acetoxyl group must be connected at C-5 in 4. Thus, the structure of 4 was determined as 5-hydroxy-4-oxopentyl acetate and named catathelasmol D, which was also a new natural product,¹¹ as shown in Figure 2.

Catathelasmol E (5), a colorless oil, was assigned the molecular formula $C_9H_{16}O_5$ by the positive-ion HR-ESI-MS: 227.0889 (calcd. for $C_9H_{16}O_5$ Na, 227.0895). The NMR data of 5 were similar to those of 3, and comparison of ¹³C-NMR data showed that instead of a ketone, one oxymethine carbon at δ 69.3 (d) newly appeared in 5. The obvious differences in the ¹H-NMR spectra were as follows: (a) the oxymethylene singlet in 3 was observed separately from each other at δ 3.95 (1H, dd, *J*=11.3, 7.2 Hz) and 4.11 (1H, dd, *J*=11.3, 7.2 Hz),



Figure 3 Chemical transformation from 4 to 1.

respectively in 5; and (b) an oxygen-bearing methine proton at δ 3.84 (1H, m) was newly detected in 5. The above NMR character indicated that the keto group in 3 was hydrogenated in 5. Initially, the modified Mosher's method^{12,13} was applied to determine the absolute configuration of the hydroxyl at C-2, but failed. Alternatively, we resorted to the computational method to establish the absolute configuration. The absolute configuration of 5 was assigned as (*R*) using matrix model.¹⁴ The det(*D*) value for this chiral alcohol was -2.50 and the calculated k_0 value was 4.2, which is located in the window of coefficients of chiral secondary alcohols in chloroform. Accordingly, the structure of 5 was established as 2-hydroxypentane-1,5-diyl diacetate, shown in Figure 2, and was named catathelasmol E.

It was not until quite recently that such simple pentanol derivatives were found, which suggested that this type of metabolites were distributed in a narrow range, so we believe that they may possess an important chemotaxonomic significance. Besides, the realization of the chemical transformation from 4 to 1 validated the structural correctness (Figure 3).

Chemical transformation from 4 to 1:¹⁵ A solution of 4 (50 mg) in MeOH (5.0 ml) was treated with K₂CO₃ (97.4 mg) at room temperature. After stirring for 1.5 h, saturated aqueous ammonium chloride was added and extracted with ethyl acetate, and the organic layer was washed with water and brine, dried over sodium sulfate, and evaporated to a residue that was purified by a silica gel column using pure CHCl₃ to yield 1 (12 mg). The determined optical rotation was expectedly zero. Thus, the cyclization of 4 to 1 obtained in the experiments must involve the enzyme catalysis.

Glucocorticoid hormones play important roles in many biological and physiological processes, including regulation of energy metabolism, inflammatory, immune and stress responses, and cardiovascular homeostasis. The action of glucocorticoid on target tissue is not dependent inevitably on the circulating levels, but is regulated in a tissue-specific manner by the enzymes of 11β-hydroxysteroid dehydrogenases (11 β -HSD1 and 11 β -HSD2), which catalyze the interconversion of active 11-hydroxy-glucocorticoids (cortisol in humans and corticosterone in rodents) and their respective inert 11-keto forms (cortisone in humans and 11-dehydrocorticosterone in rodents).¹⁶ 11β-HSD1 is highly expressed in the liver, gonad, adipose tissue and brain, in which it acts as a reductase regenerating the active glucocorticoids from its inactive forms, thus amplifying local glucocorticoid action.17 11β-HSD2 is predominantly expressed in aldosterone target cells, such as the kidney and colon, in which it catalyzes the inactivation of glucocorticoids, thereby preventing the excessive activation of the mineralocorticoid receptor and sequelae, including sodium retention, hypokalemia and hypertension.

We tested the inhibitory effect of the compounds on both human and mouse 11 β -HSD1 and 11 β -HSD2. Compound **3** showed inhibitory activities against 11 β -HSD1 (human IC₅₀=28.7 µg ml⁻¹; mouse IC₅₀=30.4 µg ml⁻¹) and 11 β -HSD2 (human IC₅₀=5.1 µg ml⁻¹; mouse IC₅₀=32.3 μg ml⁻¹). Compound **4** showed inhibitory activities against 11β-HSD1 (human IC₅₀=47.4 μg ml⁻¹; mouse IC₅₀= 149.2 μg ml⁻¹) and 11β-HSD2 (human IC₅₀=38.9 μg ml⁻¹; mouse IC₅₀=129.1 μg ml⁻¹). Compound **5** showed inhibitory activities against human 11β-HSD1 (IC₅₀=62.3 μg ml⁻¹) and 11β-HSD2 (IC₅₀=177.0 μg ml⁻¹). Therefore, compounds **3**–5 showed inhibitory activities against 11β-HSD1 and 11β-HSD2 and provide the possibility for modulating local cortisone/cortisol availability *in vivo*.

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New atpenins, NBRI23477 A and B, inhibit the growth of human prostate cancer cells

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The growth and metastasis of prostate cancer are regulated by prostate stroma through the tumor-stromal cell interactions. Small molecules that modulate the tumor-stromal cell interactions will be new anticancer drugs. In the course of our screening of the modulators, we isolated two new atpenins, NBRI23477 A (4) and B (5), from the fermentation broth of *Penicillium atramentosum* PF1420. Compounds 4 and 5 as well as atpenin A4 (1), A5 (2) and B (3) inhibited the growth of human prostate cancer DU-145 cells in the coculture with human prostate stromal cells more strongly than that of DU-145 cells alone. *The Journal of Antibiotics* (2009) **62**, 243–246; doi:10.1038/ja.2009.20; published online 13 March 2009

Keywords: antitumor drug; atpenin; prostate cancer; prostate stroma

INTRODUCTION

The growth and metastasis of prostate cancer are regulated by prostate stroma.^{1,2} We have reported earlier that prostate stromal cell (PrSC) promotes the growth of human prostate cancer cells through the secretion of insulin-like growth factor-I.^{3,4} There is a possibility that small molecules could inhibit cancer cell growth by modulating tumor–stromal cell interactions. We developed the *in vitro* coculture system of human prostate cancer cells and PrSC, in which the growth of prostate cancer cell is increased by the coculture with PrSC.^{3,5} Using the assay method, we have been searching for the modulators of the tumor–stromal cell interactions. In the course of our screening of the modulators, we isolated new atpenins, NBRI23477 A (4) and B (5), along with the known compounds, atpenin A4 (1), A5 (2) and B (3).^{6,7} Here we describe the isolation, structure determination and biological activity of 4 and 5. We also report the activity of 1, 2 and 3 on our assay.

MATERIALS AND METHODS

Reagents

Rhodanile blue was purchased from Aldrich (Milwaukee, WI, USA). Insulin and hydrocortisone were obtained from Sigma (St Louis, MO, USA). Transferrin was obtained from Wako Pure Chemical Industries (Tokyo, Japan). The recombinant human basic fibroblast growth factor was purchased from Pepro Tech (London, UK).

Cells

The human prostate cancer DU-145 cells were obtained from American Type Culture Collection (Manassas, VA, USA) and maintained in Dulbecco's modified Eagle's medium supplemented with 10% fetal bovine serum (ICN Biomedicals, Aurora, OH, USA), 100 U ml⁻¹ penicillin G and 100 μ g ml⁻¹ streptomycin at 37 °C with 5% CO₂. The human normal PrSCs were obtained

from Bio Whittaker (Walkersville, MD, USA) and maintained in Dulbecco's modified Eagle's medium supplemented with 10% fetal bovine serum, $100\,U\,ml^{-1}$ penicillin G, $100\,\mu g\,ml^{-1}$ streptomycin, ITH ($5\,\mu g\,ml^{-1}$ insulin, $5\,\mu g\,ml^{-1}$ transferrin and $1.4\,\mu M$ hydrocortisone) and $5\,n g\,ml^{-1}$ human basic fibroblast growth factor at 37 °C with 5% CO₂.

Coculture experiment

A microplate assay method for the selective measurement of epithelial tumor cells in coculture with stromal cells using rhodanile blue dye was performed as described before.⁵ PrSCs were first inoculated into 96-well plates at 5000 cells per well in 100 μ l of Dulbecco's modified Eagle's medium supplemented with ITH and 0.1% fetal bovine serum in the presence of various concentrations of the test compounds. After 2 days, 10 μ l of DU-145 cell suspension (5000 cells) in serum-free Dulbecco's modified Eagle's medium was inoculated onto a monolayer of PrSC, and the cells were further cultured for 3 days. For monoculture of DU-145 cells, the assay medium alone was first incubated in the presence of test compounds for 2 days at 37 °C. Then, DU-145 cells were inoculated as described above and cultured for further 3 days.

Analytical measurement

Melting points were obtained on a Yanagimoto micro melting point apparatus (Yanagimoto, Kyoto, Japan). Optical rotations were measured on a JASCO P-1030 polarimeter (JASCO, Tokyo, Japan). UV spectra were recorded on a Hitachi 228 A spectrometer (Hitachi, Tokyo, Japan). ¹H- and ¹³C-NMR spectra were measured on a JEOL JNM A400 spectrometer (JEOL, Tokyo, Japan) using TMS as an internal standard. High resolution electrospray ionization mass spectrometer (JEOL).

Fermentation of fungal strain PF1420

Penicillium atramentosum PF1420 was isolated from a soil sample collected in Iwamizawa, Hokkaido, Japan. A slant culture of *P. atramentosum* PF1420 was used to inoculate 100-ml Erlenmeyer flasks. Each contained 20 ml of a seed

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medium consisting of 2.0% soluble starch, 1.0% glucose, 0.2% soybean meal, 0.6% wheat germ, 0.5% polypeptone, 0.3% yeast extract and 0.2% CaCO3 in deionized water adjusted to pH 7.2 with NaOH solution before sterilization. The flasks were incubated at 25 °C for 72 h on a rotary shaker at 220 r.p.m. Portions of 1.0 ml of this seed culture were transferred into six 500-ml Erlenmeyer flasks, each of which contained 100 ml of a seed medium. The flasks were incubated at 25 °C for 48 h on a rotary shaker at 220 r.p.m. Portions of 150 ml of this seed culture were transferred into four stainless vats, each of which contained 2.5% soybean meal and water-absorbed rice (4 kg) as solid production medium. The stainless vats were thoroughly stirred and then statically cultured at 25 °C for 14 days. After incubation, 16-kg portion of the obtained culture was extracted with 321 of 67% aqueous acetone.

RESULTS

Isolation procedure for atpenins

The 16-kg culture broth of P. atramentosum PF1420 was extracted with 321 of 67% aqueous acetone. The filtrate of the extracts was concentrated in vacuo to remove acetone. The aqueous solution (51, pH 7) was applied on an HP-20 column. After washing the column with H₂O and 50% MeOH, active ingredients were eluted with 100% MeOH. The eluate was concentrated in vacuo, dissolved in 600 ml H₂O and then extracted with EtOAc. The organic layer was dried over Na₂SO₄ and concentrated *in vacuo* to afford 9.92 g of dried material. The materials were applied on a silica gel column (450 g, Wakogel C-200, 75-150 µm; Wako, Osaka, Japan) prepared with CHCl₃, and eluted with CHCl3 and CHCl3-MeOH. The fractions eluted with CHCl₃-MeOH (25:1) were concentrated in vacuo to give 4.13 g of viscous material. The viscous material was applied on gel filtration chromatography of Sephadex LH-20 (MeOH). The fractions containing atpenins were concentrated in vacuo to give 2.76g of crude material. The crude material was purified by a reversed-phase HPLC column (Inertsil ODS-3, 20×250 mm, 6.0 ml min^{-1}) with 50% CH₃CN to afford crude atpenins, 46.4 mg of 1, 119.8 mg of 2,

Table 1 Physicochemical properties of 4 and 5

White powder

130-132 °C

C15H21NO5CI2

364.0713 (M-H)-

236. 273. 329

_37

364.0719 for $C_{15}H_{20}NO_5CI_2$

2935, 1651, 1597, 1448,

1325, 1198, 1163, 995

4

111.1 mg of 3 and 4, and 14.5 mg of 5. The crude sample of 5 was applied on a Sephadex LH-20 column (MeOH) to afford pure 8.0 mg of 5. The crude mixture of 3 and 4 was further purified by a reversedphase HPLC column (Inertsil ODS-3, 10×250 mm, 3.0 ml min⁻¹) with 75% CH₃CN and 0.1% TFA, and then the fractions containing

Table 2 The ¹³C- and ¹H-NMR assignments of 4 and 5 in pyridine-d₅

		4	5			
Position	¹³ C p.p.m. (mult.)	¹ H p.p.m. (mult., J(Hz))	¹³ C p.p.m. (mult.)	¹ H p.p.m. (mult., J(Hz))		
2 ^a	162.7 (s)		162.8 (s)			
3	100.6 (s)		101.0 (s)			
4 ^a	165.9 (s)		165.8 (s)			
5	125.0 (s)		124.9 (s)			
5-0CH ₃	60.7 (q)	3.84 (s)	60.7 (q)	3.83 (s)		
6	160.2 (s)		160.1 (s)			
6-0CH ₃	54.4 (q)	3.90 (s)	54.4 (q)	3.89 (s)		
1′	210.6 (s)		211.3 (s)			
2′	42.2 (d)	4.36 (m)	42.2 (d)	4.42 (m)		
2'-CH3	16.4 (q)	1.35 (d, 6.4)	18.1 (q)	1.33 (d, 6.8)		
3′	36.8 (t)	2.06-2.13 (m)	41.0 (t)	1.42 (m)		
				2.10 (m)		
4′	48.0 (d)	2.40 (m)	36.6 (d)	2.33 (m)		
4'-CH ₃	15.8 (q)	1.34 (d, 6.4)	20.5 (q)	1.07 (d, 6.8)		
5′	97.9 (s)		145.2 (d)	5.79 (ddd, 7.6, 10.4, 17.2)		
6′	35.7 (q)	2.14 (s)	114.4 (t)	4.95 (dd, 1.2, 10.4) 4.99 (dd, 1.2, 17.2)		

^aAssignments may be interchanged.

Chemical shifts in p.p.m. from TMS as an internal standard.

The ¹³C- and ¹H-NMR were measured at 100 and 400 MHz, respectively.



NBRI23477 B (5)

Figure 2 Partial structure of 4.



NBRI23477 A (4)

Atpenin A4 (1): R₁=Cl, R₂=H Atpenin A5 (2): R1=CI, R2=CI Atpenin B (3): R₁=H, R₂=H

5

Hygroscopic solid

294.1361 (M-H)

235. 271. 331

-39°

294.1342 for C15H20NO5

1325, 1196, 1163, 995

C15H21NO5

Figure 1 Structures of 1-5.

Appearance

Found

Calcd

Melting point

Molecular formula

HR-ESI-MS (m/z)

 $UV\lambda_{max}$ (nm) (MeOH)

 $[\alpha]_{D}^{20}(c \ 0.2, \text{ EtOH})$

IRvmax (KBr) (cm⁻¹)

244

245

3 and **4** were applied on a Sephadex LH-20 column (MeOH) to afford pure 72.4 mg of **3** and 4.5 mg of **4**, respectively. The crude mixture of **1** and **2** was further purified by a reversed-phase HPLC column (Inertsil ODS-3, 10×250 mm, 3.0 ml min⁻¹) with 60% CH₃CN and 0.1% TFA, and then the fractions containing **1** and **2** were applied on a Sephadex LH-20 column (MeOH) to afford pure 15.5 mg of **1** and 26.1 mg of **2**, respectively. Separation by analytical HPLC (Inertsil ODS-3, 3μ m, 4.6×150 mm, 1.0 ml min⁻¹) with 75% CH₃CN and 0.1% TFA gave the following retention times (in minutes): 4.38 (1), 4.40 (2), 5.29 (3), 4.95 (4) and 4.28 (5).

Physicochemical properties

The physicochemical properties of **4** and **5** are summarized in Table 1. They are soluble in MeOH and DMSO. The molecular formulae of **4** and **5** were determined to be $C_{15}H_{21}NO_5Cl_2$ and $C_{15}H_{21}NO_5$, respectively, by HR-ESI-MS. The general features of their UV and NMR spectra resembled each other, indicating structural similarities of these compounds. Compounds **1–3** were identified by NMR spectra as atpenin A4 (1), A5 (2) and B (3), respectively (Figure 1).^{6,7}

Structure determination of NBRI23477 A (4)

The ¹H- and ¹³C-NMR data of 4 (Table 2) were similar to those of 2.⁸ However, the signals of 5'-methine and 6'-methylene of 2 were not observed in the ¹³C-NMR spectrum of 4, but a quaternary carbon (δ 97.9) and a methyl carbon (δ 35.7) appeared in 4. The partial structure of 4 was established by analyses of ¹H-¹H correlation spectroscopy (COSY) and heteronuclear multiple bond connectivity (HMBC) spectra (Figure 2). ¹H-¹H COSY spectrum revealed the following fragment: -CH(CH₃)-CH₂-CH(CH₃)-. In the HMBC spectrum, singlet methyl protons (6'-H, δ 2.14) correlated to a quaternary carbon (C-5', δ 97.9) and a methine carbon (C-4', δ 48.0). Methyl protons (δ 1.35) connecting to C-2' methine correlated to a carbonyl carbon (C-1', δ 210.6). Therefore, the total structure of **4** was found to be a new family of atpenin (Figure 1).

Structure determination of NBRI23477 B (5)

The ¹³C- and ¹H-NMR data of **5** (Table 2) were similar to those of **3**.^{6,7} However, the signals of 5'-methylene and 6'-methyl of **3** were not observed in the ¹³C-NMR spectrum of **5**. On the other hand, an olefine carbon (δ 145.2) and a terminal olefine carbon (δ 114.4) were observed in **5**. The partial structure of **5** was established by analyses of the ¹H-¹H COSY and HMBC spectra (Figure 3). ¹H-¹H COSY



Figure 3 Partial structure of 5.



Figure 4 Effects of 1–5 on coculture of DU-145 cells and PrSC. The growth of DU-145 cells cocultured with PrSC (\bullet) or that of DU-145 cells alone (\bigcirc) in the presence of the indicated concentrations of 1–5 was determined using rhodanile blue method. Values are means of duplicate determinations. Each s.e. is less than 10%.

spectrum revealed the following fragment: $-CH(CH_3)-CH_2-CH(CH_3)-CH=CH_2$. In the HMBC spectrum, a methyl proton (δ 1.33) connecting to C-2' methine correlated to a carbonyl carbon (C-1', δ 211.3). Therefore, the total structure of **5** was found to be a new family of atpenin (Figure 1).

Biological activities

The effects of 1–5 on coculture of human prostate cancer DU-145 cells with PrSC were determined using rhodanile blue staining method.⁵ In the coculture, the growth of DU-145 cells is increased by PrSC.^{3,5} As shown in Figure 4, all compounds showed selective growth inhibitory activities and inhibited the growth of DU-145 cells in coculture with PrSC more strongly than that of DU-145 cells alone. The IC₅₀ values of 1–5 against the growth of DU-145 cells in coculture were 0.21, 0.021, 0.034, 0.064 and 0.13 µg ml⁻¹, respectively, whereas those of 1–5 against the growth of DU-145 cells alone were 0.85, 0.048, 0.54, 0.95 and 0.71 µg ml⁻¹, respectively. All compounds did not show apparent cytotoxicity against stromal cells under microscopic observation (data not shown).

DISCUSSION

In this study, we have also obtained three structurally related compounds in addition to 1–5. The HR-ESI-MS spectra revealed that the molecular formulae of these compounds were $C_{15}H_{22}NO_6Cl$, $C_{15}H_{23}NO_6$ and $C_{15}H_{20}NO_5Cl_3$, respectively. Among them, a compound having the molecular formula of $C_{15}H_{20}NO_5Cl_3$ may be identical to reported WF-16775 A2,⁸ but we could not elucidate the structures of additional three compounds further due to their trace amounts. Ōmura *et al.*⁶ reported that there were atpenins A1, A2 and A3 along with A4, A5 and B, but they did not obtain A1, A2 and A3 in pure form and did not show any structural information. We cannot exclude the possibility that **4** and **5** would be identical to one of them. However, we have actually presented here two new structures of atpenins.

Atpenins A4, A5 and B were originally isolated as antifungal antibiotics.^{6,7} Thereafter, atpenin B was found to decrease the cellular adenosine 5'-triphosphate.⁹ Furthermore, it is reported that atpenins

specifically inhibit mitochondrial complex II (succinate–ubiquinone oxidoreductase).¹⁰ Mitochondria is now considered as a rational target for cancer therapy.¹¹ Although there is a possibility that atpenins modulate tumor–stromal cell interactions by inhibiting mitochondrial functions, the elucidation of the precise mechanism of action needs to be studied further. We are now studying the effects of atpenins on tumor growth *in vivo* using mouse xenograft models.

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Role of the AraC–XyIS family regulator YdeO in multi-drug resistance of *Escherichia coli*

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Multi-drug efflux pumps contribute to the resistance of *Escherichia coli* to many antibiotics and biocides. In this study, we report that the AraC–XyIS family regulator YdeO increases the multi-drug resistance of *E. coli* through activation of the MdtEF efflux pump. Screening of random fragments of genomic DNA for their ability to increase β-lactam resistance led to the isolation of a plasmid containing *ydeO*, which codes for the regulator of acid resistance. When overexpressed, *ydeO* significantly increased the resistance of the *E. coli* strain to oxacillin, cloxacillin, nafcillin, erythromycin, rhodamine 6G and sodium dodecyl sulfate. The increase in drug resistance caused by *ydeO* overexpression was completely suppressed by deleting the multifunctional outer membrane channel gene *tolC*. TolC interacts with different drug efflux pumps. Quantitative real-time PCR showed that YdeO activated only *mdtEF* expression and none of the other drug efflux pumps in *E. coli*. Deletion of *mdtEF* completely suppressed the YdeO-mediated multi-drug resistance. YdeO enhances the MdtEF-dependent drug efflux activity in *E. coli*. Our results indicate that the YdeO regulator, in addition to its role in acid resistance, increases the multi-drug resistance of *E. coli* by activating the MdtEF multi-drug efflux pump.

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Keywords: drug efflux pump; Escherichia coli; MdtEF; multidrug resistance; YdeO

INTRODUCTION

Multi-drug efflux pumps cause serious problems in cancer chemotherapy and in the treatment of bacterial infections. Bacterial drug resistance is often associated with multi-drug efflux pumps that decrease drug accumulation in the cell.^{1,2} Bacterial multi-drug efflux pumps are classified into five families on the basis of sequence similarity: major facilitator, resistance-nodulation-cell division (RND), small multi-drug resistance, multidrug and toxic compound extrusion, and ATP-binding cassette.3-5 Of these, RND family efflux pumps play major roles in both intrinsic and elevated resistance of Gram-negative bacteria to a wide range of compounds, including β-lactams.^{1,6-12} RND efflux pumps require two other proteins to function: a membrane fusion protein and an outer membrane protein. Many drug efflux pumps in Escherichia coli need TolC to function.^{12–15} TolC is responsible for resistance to various antibiotics, including β-lactams,¹² quinolones¹⁶ and macrolides.¹⁷ Bacterial genome sequencing enables us to trace drug-resistance genes.¹⁸⁻²⁰ There are many putative and proven drug efflux pumps in the E. coli genome, and we have identified earlier 20 functional drug efflux pumps.^{9,20} As many such efflux pumps have overlapping substrate spectra,⁹ it is intriguing that bacteria, with their economically organized genomes, harbor such large sets of multi-drug efflux genes.

The key to understanding how bacteria utilize these multiple efflux pumps lies in the regulation of pump expression. The currently available data show that multi-drug efflux pumps are often expressed under precise and elaborate transcriptional control.^{21–24} Expression of *acrAB*, which encodes the major AcrAB efflux pump, is subject to multiple levels of regulation. In *E. coli*, it is modulated locally by the repressor AcrR²⁵ and AcrS.²⁶ At a more global level, it is modulated by stress conditions and by global regulators, such as MarA, SoxS and Rob.^{27,28} These examples illustrate the complexity and diversity of the mechanisms regulating bacterial multi-drug efflux pumps.

Stomach acid (pH ≤ 2) kills most bacteria;²⁹ however, *E. coli* survives this acidity.^{30,31} The acid resistance appears to contribute to the low infectious dose of pathogenic *E. coli*, and aids in the gastric passage of commensal strains.^{32,33} One of the most efficient acid resistance systems in *E. coli*, the Gad system, is based on the coordinated action of two isoforms of glutamate decarboxylase (GadA and GadB) and of a specific glutamate/ γ -aminobutyrate (GABA) antiporter (GadC).^{34,35} The *gadA/BC* genes, activated in response to acid stress, are subject to complex circuits of regulation involving the AraC–XylS family regulator YdeO.³⁶ In this study, we demonstrate that YdeO contributes to the multi-drug resistance as well as the acid resistance of *E. coli*. The results suggest a role of YdeO in the multi-drug resistance of *E. coli*.

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MATERIALS AND METHODS

Bacterial strains, plasmids and growth conditions

The bacterial strains and plasmids used in this study are listed in Table 1. The *E. coli* strains were derived from the wild-type strain MG1655.³⁷ Phage P1mediated transductions were performed as described earlier.³⁸ Bacterial strains were grown at 37 °C in Luria–Bertani (LB) broth.³⁹ Cells were collected for total RNA extraction when the cultures reached an optical density of 0.6 at 600 nm.

Screening for positive regulators of multi-drug resistance

DNA manipulation generally followed standard practice.³⁹ A genomic library was developed by partial *Sau*3AI digestion of the chromosomal DNA as follows. Chromosomal DNA prepared from an overnight culture of the wild-type strain MG1655³⁷ was digested with *Sau*3AI ($1 \cup \mu l^{-1}$) for 15, 20, 30 and 40 min. The digested DNA was separated on a 0.8% agarose gel, and fragments approximately 0.5–3 kb in size were purified and ligated into the *Bam*HI site of vector pHSG398 (Takara Bio Inc., Otsu, Japan). The ligation products were transformed into *E. coli* DH5 α^{39} to select chloramphenicol-resistant transformants. Plasmid DNA was prepared from a pool of 16,000 transformants and used to transform the *acrB* deletion strain NKE96. Cells were plated on LB agar medium³⁹ containing 15 µg ml⁻¹ chloramphenicol and inhibitory concentrations of various drugs.

Plasmid construction

The *ydeO* gene was amplified from MG1655 genomic DNA using the primers *ydeO*-F_BamHI and *ydeO*-R_SalI listed in Table 2, which introduced BamHI and SalI sites at the ends of the amplified fragment. This PCR product was cloned between the BamHI and SalI sites of the vector pHSG398 (Takara Bio Inc.) to produce the plasmid pydeO.

Construction of gene deletion mutants

Gene deletion was performed according to the method of Datsenko and Wanner, with recombination between short homologous DNA regions catalyzed by phage λ Red recombinase.⁴⁰ A curable expression plasmid encoding Red recombinase (pKD46) was introduced into the MG1655 strain. The chloramphenicol resistance gene *cat* or the kanamycin resistance gene *kan*, flanked by Flp recognition target sites, was amplified by PCR using the primers listed in Table 2. The plasmid pKD3 or pKD4 was used as a template. This PCR product was used to transform the MG1655 strain expressing Red recombinase, and recombinant clones were isolated as chloramphenicol- or kanamycin-

resistant colonies. The pKD46 vector was eliminated by incubating at a nonpermissive temperature (37 °C), as confirmed by the loss of ampicillin resistance. The chromosomal structure of the mutated loci was verified by PCR, as described earlier,⁴⁰ and by Southern hybridization using probes specific for (i) the antibiotic resistance genes used during the construction of chromosomal deletions and (ii) sequences flanking the inactivated loci. The deletions were then transferred to the wild-type MG1655 strain by P1 transduction. The *cat* and *kan* genes were eliminated using the plasmid pCP20, as described earlier.⁴⁰

Determination of the MICs for toxic compounds

The antibacterial activities of different agents were determined on Luria agar (1% tryptone, 0.5% yeast extract and 0.5% NaCl) plates containing various concentrations of the compounds (Sigma-Aldrich, St Louis, MO, USA) listed in Table 3. The agar plates were prepared by the two-fold agar dilution technique, as described earlier.⁴¹ The MIC was defined as the lowest concentration of a compound that inhibited cell growth. To determine the MICs, bacteria were grown in LB broth at 37 °C overnight and diluted in the same medium. Then, the organisms were tested at a final inoculum size of $10^4 \, \text{cfu} \, \mu \text{l}^{-1}$ using a multipoint inoculator (Sakuma Seisakusyo, Tokyo, Japan), and were incubated at 37 °C for 20 h.

RNA extraction

Total RNA was isolated from bacterial cultures using the RNeasy Protect Bacteria Mini Kit (Qiagen, Hilden, Germany) and RNase-Free DNase (Qiagen), as described earlier.⁴² The total RNA was isolated from exponential-phase cultures of *ΔacrB*/vector (NKE154) and *ΔacrB*/pydeO (NKE169). The absence of genomic DNA from the DNase-treated RNA samples was confirmed by both non-denaturing agarose electrophoresis gels and PCR with primers against target genomic DNA. The RNA concentration was then determined spectro-photometrically.³⁹

Determination of specific transcript levels by quantitative real-time PCR following reverse transcription (RT)

Bulk cDNA samples were synthesized from total RNA using TaqMan Reverse Transcription Reagents (Applied Biosystems, Carlsbad, CA, USA) and random hexamers, as described earlier.^{43,44} The specific primer pairs listed in Table 2 were designed using ABI PRISM Primer Express software (Applied Biosystems). *rrsA* of 16S rRNA was chosen as the normalizing gene.⁴⁵ Real-time PCR was

Strains or plasmids	Original names	Characteristics	Sources or references
Strains as in text			
WT	MG1655	Wild type	37
$\Delta a cr B$	NKE 96	$\Delta a cr B$	Present study
$\Delta a cr B$ /vector	NKE 154	∆ <i>acrB</i> /pHSG398	Present study
$\Delta a cr B/py de O$	NKE 169	$\Delta acrB/pydeO$	Present study
Δ acrB tolC	NKE 128	$\Delta acr B \Delta tol C$	Present study
∆acrB tolC/vector	NKE 160	ΔacrB ΔtolC/pHSG398	Present study
∆acrB tolC/pydeO	NKE 174	$\Delta acr B \Delta tol C/pydeO$	Present study
Δ acrB mdtEF	NKE 139	$\Delta acr B \Delta m dt EF$::Km ^R	Present study
$\Delta a crB mdtEF$ /vector	NKE 176	∆acrB ∆mdtEF::Km ^R /pHSG398	Present study
$\Delta a crB mdtEF/pydeO$	NKE 180	$\Delta acrB \Delta mdtEF$::Km ^R /pydeO	Present study
Plasmids			
pKD46		Red recombinase expression plasmid, Ap ^R	40
pKD3		rep _{R6Kv[0]} Ap ^R FRT Cm ^R FRT	40
pKD4		rep _{R6Kv[0]} Ap ^R FRT Km ^R FRT	40
pCP20		rep_{pSC101} ts Ap ^R Cm ^R c/857 λ P _R f/p	40
pHSG398		rep _{pMBI} Cm ^R	Takara Bio Inc.
p <i>ydeO</i>		ydeO gene cloned into pHSG398, Cm ^R	Present study

Table 2 Primers used in the present study

Primers	Sequences (5'-3')
For gene deletion	
to/C-P1	ACTGGTGCCGGGCTATCAGGCGCATAACCATCAGCAATAGGTGTAGGCTGGAGCTGCTTC
toIC-P2	TTACAGTTTGATCGCGCTAAATACTGCTTCACCACAAGGACATATGAATATCCTCCTTAG
acrB-P1	AAAAAGGCCGCTTACGCGGCCTTAGTGATTACACGTTGTAGTGTAGGCTGGAGCTGCTTC
acrB-P2	GAACAGTCCAAGTCTTAACTTAAACAGGAGCCGTTAAGACCATATGAATATCCTCCTTAG
mdtF_P1	
mate P2	
For gene cloning	
<i>ydeO</i> -F_BamHI	CGCGGATCCAAACAACAGCAAATTATAAA
ydeO-R_Sall	CGCGTCGACTTTCAATAAATATATCGCCTA
For quantitative PCR	
rrsA-F	CGGTGGAGCATGTGGTTTAA
rrsA-R	GAAAACTTCCGTGGATGTCAAGA
acrA-F	GTCTATCACCCTACGCGCTATCTT
acrA-R	GCGCGCACGAACATACC
acrD_F	GTACCCTGGCGATTTTTCATT
acrDR	
	CATCATTECCECAAAACC
	TTEECECAETEAETTEETA
acrE-N	
ber P	
CUSB-F	
<i>cusb</i> -R	
emrA-F	GCGAAIAIIGAGGIGCAGAAAA
<i>emrA</i> -R	GGCACACGGCGGTTGTA
emrD-F	GTGGATCCCCGACTGGTTT
<i>emrD-</i> R	CCCGGCACCGAAAAAGA
emrE-F	GGTATTGTCCTGATTAGCTTACTGTCAT
<i>emrE-</i> R	GCACAAATCAACATCATGCCTATAA
emrK-F	GCGCTTAAACGTACGGATATTAAGA
<i>emrK</i> -R	ACTGTTTCGCCGACCTGAAC
<i>fsr</i> -F	TGGTGTTGGCGCAAATCA
<i>fsr</i> -R	TCGTCGCTTTGGGTTTTCC
macA-F	CGGTGATTGCCGCACAA
macA-R	TTACCAGCATGGCGCTCAT
mdfA-F	CTTGCTGTTAGCGCGTCTGA
mdfA-R	GCCAGCCGCCCATAATAAT
mdtA-F	CGCCGTAGAACAGGCAGTTC
<i>mdtA</i> -R	TGCGCACCGTAACGGTATTA
mdtE-F	CCCCCGGTTCGGTCAA
<i>mdtE</i> -R	GGACGTATCTCGGCAACTTCAT
mdtF-F	TTACCGTCAGCGCTACCTATCC
mdtF-R	GCCATCAAGCCCATTCATATTT
mdtG-F	CGGTATTGTCTTCAGCATTACATTTT
mata i	GGCGAGTCCACCCCAAA
mata n mdtH_F	TTTTCACCCTGATTTGTCTGTTTTAT
mdtH P	
mdt / E	
muu-i mat I P	
mail-t	
matL-K	
mdtM-F	CGIGATTITAAIGCCGAIGICA
<i>mdtM</i> -R	GCCATACCGCCAGCAAGAT
<i>toIC</i> -F	CCGGGATTTCTGACACCTCTT
to/C-R	TTTGTTCTGGCCCATATTGCT

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Table 3 Susceptibility of *E. coli* strains to β -lactams and toxic compounds

	MIC (μg/ml)					
Strains	OXA	MCIPC	NAF	ERM	R6G	SDS
WT	256	>512	512	128	>512	>512
$\Delta a cr B$	0.5	1	2	4	2	64
∆ <i>acrB</i> /vector	0.5	1	2	4	2	64
∆ <i>acrB</i> /p <i>ydeO</i>	16	64	32	64	256	>512
$\Delta a crB to IC$	0.5	0.25	1	2	2	16
∆acrB tolC/vector	0.5	0.25	1	2	2	16
∆acrB tolC/pydeO	0.5	0.25	1	2	2	16
$\Delta a cr B m dt EF$	0.5	1	2	4	2	64
∆acrB mdtEF/vector	0.5	1	2	4	2	64
∆acrB mdtEF/pydeO	0.5	1	2	4	2	64

Abbreviations: ERM, erythromycin; MCIPC, cloxacillin; NAF, nafcillin; OXA, oxacillin; R6G, rhodamine 6G; SDS, sodium dodecyl sulfate. Values in bold face are larger than those of a corresponding parental strain haboring the

Values in bold face are larger than those of a corresponding parental strain haboring to pHSG398 vector. MIC determinations were repeated at least three times.

performed with each specific primer pair using SYBR Green PCR Master Mix (Applied Biosystems). The reactions were run on an ABI PRISM 7000 Sequence Detection System (Applied Biosystems); the fluorescence signal due to SYBR Green intercalation was monitored to quantify the double-stranded DNA product formed in each PCR cycle. The expression levels of drug efflux pump genes and *tolC* in *AacrB/pydeO* (NKE169) were compared with those in *AacrB/* vector (NKE154).

Acid resistance assay

A single colony of an E. coli strain harboring plasmid was inoculated into 1 ml of LB broth containing chloramphenicol and grown overnight at 37 °C. The LB broth (20 ml) was inoculated with 0.1 ml of the overnight culture and grown at 37 °C. When the cultures reached a cell density of 2×10^8 cfu ml⁻¹, 50 µl of the culture was transferred to 2 ml of phosphate-buffered saline (PBS; pH 7.2) and to 2 ml of warmed LB broth (pH 2.5, adjusted with HCl). The cfu ml⁻¹ in PBS was determined by plating serial dilutions in PBS buffer (pH 7.2) on LB agar and using these as initial cell populations. The LB broth (pH 2.5) inoculated with E. coli was incubated at 37 °C for 1 h, and the cfu ml⁻¹ in the LB broth (pH 2.5) was determined as described above and used as the final cell population. The percentage acid survival was then calculated as the number of cfu ml⁻¹ remaining after acid treatment divided by the initial cfu ml⁻¹ at time zero. Each experiment was performed in triplicate. The percentage survival values were converted to logarithmic values ($\log_{10} x$, where x equals the percentage survival) for the calculation of geometric means and standard errors (s.e.).

Drug efflux assay

The drug efflux activities of *E. coli* cells were measured using cells preloaded with rhodamine 6G. The exponential cultures of *E. coli* cells were harvested and washed twice with 100 mM potassium phosphate buffer (pH 7.5) containing 5 mM MgSO₄. For maximal accumulation of the fluorophore, the cells (optical density of 1.0 at 600 nm) were incubated with 1 μ M rhodamine 6G and 40 μ M carbonyl cyanide *m*-chlorophenylhydrazone (CCCP) at 37 °C for 1 h. The cells were then centrifuged, resuspended in the same medium with the addition of 25 mM glucose as an energy source, and subjected to fluorescence measurement. The fluorescence of the compound was continuously monitored using a Hitachi model F-2000 fluorescence spectrophotometer (Hitachi High-Technologies Corp., Tokyo, Japan). Rhodamine 6G transport was measured with excitation at 529 nm and emission at 553 nm.

RESULTS

Overexpression of ydeO increases resistance to oxacillin

The expression of multi-drug efflux genes is often regulated in a complex manner, as described in the Introduction section. We there-

fore screened the genomic library of the *E. coli* for genes that increased multi-drug resistance levels in this organism. We screened a host strain lacking a functional *acrB* gene in order to identify regulatory elements involved in the expression of other multi-drug resistance systems. The library was developed from the chromosomal DNA of the MG1655 strain, and then, the recombinant plasmids were transformed into the $\Delta acrB$ strain NKE96 as described in the Materials and methods section. In one experiment, we found a 32-fold increase in oxacillin MIC against the transformant (data not shown). Introduction of the plasmid isolated from this strain into fresh $\Delta acrB$ cells resulted in the same oxacillin resistance phenotype: the MIC increased 32-fold over the recipient strain (data not shown).

Sequencing of the plasmid revealed an insertion containing the complete coding sequence of *ydeO* and a partial sequence of *yneN*. YdeO is the AraC–XylS family regulator that controls genes involved in acid resistance, such as glutamate decarboxylase genes (*gadA* and *gadB*) and a specific glutamate/GABA antiporter (*gadC*).^{34,36} It seemed likely that overexpressed YdeO caused the transcriptional activation of genes involved in oxacillin resistance in the cells carrying this plasmid.

Full-length wild-type *ydeO* was cloned into the pHSG398 vector to obtain *pydeO* (Table 1). Oxacillin MICs for NKE96 cells harboring *pydeO* were 32 times higher (16 versus $0.5 \,\mu g \,ml^{-1}$) than that for cells harboring the pHSG398 vector (Table 3), suggesting that the YdeO regulator produced by this plasmid conferred oxacillin resistance on *E. coli*. Further experiments were therefore carried out with *pydeO*.

Overexpression of *ydeO* increases resistance to β -lactams, erythromycin, rhodamine 6G and sodium dodecyl sulfate (SDS)

Our results showed that overexpression of *ydeO* increased *E. coli* resistance to oxacillin, a β -lactam antibiotic. We therefore investigated the effect of *ydeO* overexpression on the susceptibility of *E. coli* to other β -lactams. *pydeO* also increased the resistance of $\Delta acrB$ cells to cloxacillin and nafcillin (Table 3). Various other drugs were tested, including common substrates of multi-drug efflux pumps, and we found that *pydeO* increased the resistance of the $\Delta acrB$ strain to erythromycin, rhodamine 6G and SDS (Table 3). These results indicate that the overproduced YdeO regulator induces the multi-drug resistance of *E. coli*.

Effect of *tolC* deletion on multi-drug resistance modulated by the YdeO regulator

The results described above indicate that the expression of a multidrug efflux pump may be induced by overexpression of *ydeO*. In order to determine whether YdeO-mediated multi-drug resistance is attributable to TolC-dependent drug efflux pump(s), we investigated the effect of *tolC* deletion on drug resistance in cells overexpressing *ydeO*. Deletion of *tolC* from the $\Delta acrB$ strain increased susceptibility to many antimicrobial agents and chemical compounds, including cloxacillin, nafcillin, erythromycin and SDS (Table 3), which is in good agreement with earlier reports.¹² The *tolC* deletion completely inhibited YdeOmediated multi-drug resistance (Table 3). This result indicates that YdeO-mediated multi-drug resistance is attributable to increased expression of a TolC-dependent drug efflux pump.

Determination of the amounts of drug exporter transcripts by quantitative real-time RT-PCR (qRT-PCR)

In order to determine which drug efflux pump shows increased expression when ydeO is overexpressed, we used qRT-PCR to investigate changes in the amounts of drug exporter gene mRNAs. The results are shown in Figure 1. The expression levels of mdtE and mdtF



Figure 1 YdeO activates the expression of *mdtEF* multi-drug efflux genes. The amount of transcript was determined by quantitative real-time PCR as described in the Materials and methods section. The fold increase ratio was calculated by dividing the expression level of the gene in the $\Delta acrB/pydeO$ strain by that in the $\Delta acrB/vector$ strain. Experiments were performed in triplicate and the data are represented as mean ± s.d.

were significantly increased (more than 10-fold in comparison with basal levels) by *ydeO* overexpression: 87- and 48-fold increases were observed for *mdtE* and *mdtF*, respectively. Overexpression of *ydeO* did not increase the expression of other drug exporter gene transcripts (Figure 1).

Effects of deletion of the MdtEF drug efflux pump on YdeO-mediated multi-drug resistance

To determine whether the multi-drug resistance mediated by *ydeO* overexpression is because of increased expression of *mdtEF*, we investigated the effects of deleting these genes on drug resistance levels in $\Delta acrB$ /vector and $\Delta acrB/pydeO$ (Table 3). When *mdtEF* was deleted from the $\Delta acrB$ strain, there was no change in drug resistance in the resulting strains. In the $\Delta mdtEF$ acrB strain, overexpression of *ydeO* conferred no drug resistance (Table 3). Together, these data indicate that the multi-drug resistance conferred by the YdeO regulator is because of the increased expression of *mdtEF* multi-drug efflux genes.

Effects of YdeO and MdtEF on the acid resistance of E. coli

It was reported that YdeO activates the expression of genes involved in acid resistance.³⁶ In this study, we found that YdeO induces the expression of *mdtEF* multi-drug efflux genes. To test whether MdtEF contributes to acid resistance modulated by YdeO, we measured survival percentage of the wild-type strain and the $\Delta mdtEF$ mutant harboring vector (pHSG398) or pydeO at pH 2.5 (Figure 2). Overexpression of ydeO enhances the acid resistance of both the wild-type strain and the $\Delta mdtEF$ mutant (Figure 2), indicating that the MdtEF multi-drug efflux pump is not essential for YdeO-induced acid resistance.

YdeO enhances the MdtEF-dependent drug efflux activity of *E. coli* In this study, we found that YdeO activates the expression of *mdtEF* multi-drug efflux genes and drug resistance to *E. coli* cells. *E. coli* cells with a plasmid carrying the *ydeO* gene became resistant to rhodamine 6G, a toxic dye. This compound can be detected by its fluorescence. To determine whether the *ydeO*-induced drug resistance is because of efflux of this compound from the cells, we measured the efflux activity of rhodamine 6G from the cells. As shown in Figure 3, rapid efflux of rhodamine 6G from *E. coli* $\Delta acrB/pydeO$ cells was observed as an increase in fluorescence. However, no significant efflux was observed from $\Delta acrB$ and $\Delta acrB$ mdtEF/pydeO cells (Figure 3). These results indicate that the YdeO-induced multi-drug resistance is because of the



Figure 2 Effect of YdeO and MdtEF on acid resistance of *E. coli*. Acid resistance of the wild-type strain (WT) and the $\Delta mdtEF$ mutant harboring vector (pHSG398) or pydeO were grown to mid-log phase in LB broth (pH 7.0). Cells were diluted 40-fold using LB broth (pH 2.5) and incubated for 1 h at 37 °C. Error bars represent standard errors of the mean values.

enhanced drug efflux activity of *E. coli* cells, which is caused by the increased expression of *mdtEF*.

DISCUSSION

In this study, we performed a genome-wide search for a regulator of multi-drug resistance of *E. coli* by random shotgun cloning and discovered YdeO, which up-regulates *mdtEF* expression, thereby increasing the resistance to β -lactams, erythromycin, rhodamine 6G and SDS. We initially found that the plasmid carrying *ydeO* conferred oxacillin resistance on the $\Delta acrB$ strain. Then, we investigated the susceptibility of the *ydeO*-overexpressing $\Delta acrB$ strain to various drugs, including the common substrates of multi-drug efflux pumps, and found that YdeO stimulates *E. coli* resistance to oxacillin,

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Figure 3 YdeO enhances MdtEF-dependent efflux activity in *E. coli*. Active efflux of rhodamine 6G from *E. coli* $\Delta acrB$ and $\Delta acrB$ mdtEF cells overproducing YdeO was measured as described in the Materials and methods section.



Figure 4 Model for YdeO control of multi-drug resistance and acid resistance. YdeO controls expression of *gadABCE* acid resistance genes and *mdtEF* multi-drug efflux pump genes. The results of this study show that overexpression of *ydeO* activates expression of *mdtEF* and confers multi-drug resistance on *E. coli*.

cloxacillin, nafcillin, erythromycin, rhodamine 6G and SDS (Table 3). We discovered the importance of YdeO as a drug resistance factor through induction of the multi-drug efflux gene. Earlier, we reported that the MdtEF multi-drug efflux pump requires ToIC to function.¹² YdeO-mediated multi-drug resistance was completely suppressed by

deleting the *tolC* or *mdtEF* gene (Table 3). Although nothing is known of the potential YdeO-binding consensus sequence, it was reported that YdeO has an ability to bind the intergenic region between *hdeD* and *gadE*, located upstream of *mdtEF*.³⁶ The *gadE* and *mdtEF* genes are located in tandem and they might be co-transcribed. These facts suggest that the up-regulation of *mdtEF* expression might have occurred by the YdeO-binding to the upstream region of *gadE*.

An important feature of *E. coli* pathogenesis is the organism's ability to withstand extremely acidic environments (pH 2 or lower). This acid resistance contributes to the low infectious dose of pathogenic *E. coli* species. One very efficient *E. coli* acid resistance system encompasses two isoforms of glutamate decarboxylase (*gadA* and *gadB*) and a putative glutamate/GABA antiporter (*gadC*).^{34,35} It is subject to complex controls that vary with growth medium, growth phase and growth pH. Earlier work has revealed that this system is also controlled by YdeO (Figure 4).³⁶ The YdeO protein is involved in transcriptional activation of the *gadE* gene coding for a protein that regulates *gadABC* (Figure 4).^{36,46} YdeO belongs to the AraC–XylS family of bacterial transcriptional regulators known to activate acid resistance^{36,47,48} and repress virulence.⁴⁹ We investigated the contribution of the MdtEF multi-drug efflux pump in YdeO-induced acid resistance, but MdtEF is not essential for acid resistance of *E. coli*.

In addition to the roles of YdeO in acid resistance, we found that it contributes to the multi-drug resistance of *E. coli* by activating the MdtEF multi-drug efflux pump (Figure 4). The present evidence suggests that it may enhance the resistance of *E. coli* to low pH and multiple drugs in hostile environments. Further investigation of the regulation of multi-drug efflux systems in several natural environments, such as inside hosts, is required to elucidate the biological significance of their regulatory networks. Such investigations may provide further insights into the role of multi-drug efflux systems in the physiology of the cell.

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The novel anti-*Propionibacterium acnes* compound, Sargafuran, found in the marine brown alga *Sargassum macrocarpum*

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We screened extracts of 342 species of marine algae collected from Japanese coastlines for antibacterial activity against *Propionibacterium acnes*, and found a novel antibacterial compound, which we named Sargafuran, from the MeOH extract of the marine brown alga, *Sargassum macrocarpum*. Sargafuran has low cytotoxicity, and the MIC against *P. acnes* was $15 \,\mu g \,ml^{-1}$, showing a broad antibacterial activity against Gram-positive bacteria. A time-kill study showed that Sargafuran was bactericidal and completely killed *P. acnes* at $4 \times MIC$ by lysing bacterial cells. These results suggest that Sargafuran might be useful as a lead compound to develop new types of anti-*P. acnes* substances and new skin care cosmetics to prevent or improve acne. *The Journal of Antibiotics* (2009) **62**, 259–263; doi:10.1038/ja.2009.25; published online 27 March 2009

Keywords: anti-acne; antibacterial; marine alga; Propionibacterium acnes; Sargafuran; Sargassum macrocarpum

INTRODUCTION

Acne vulgaris is a common skin disease, affecting about 70–80% of adolescents and young adults. It is a multifactorial disease of the pilosebaceous unit.¹ *Propionibacterium acnes*, a common skin organism, is most often recognized in acne vulgaris and produces a number of virulence factors. Clindamycin and erythromycin are most commonly used as topical antibiotics against *P. acnes.*²

In our laboratory, we screened the biological activities of marine algae collected from the Japanese coastline and found several bioactive compounds,^{3–6} including an antibacterial compound.⁷ Other reports show that some distinct seaweeds contain antimicrobial substances against both Gram-positive and -negative bacteria.^{8,9} Thus, marine algae are a promising bioresource to find new antibacterial compounds against *P. acnes* and to develop new natural cosmetic products to prevent acne. In this study, we screened a total of 342 species of marine algae, collected from the Japanese coastline, for activity against *P. acnes*, and found a novel anti-*P. acnes* compound from one of these algae.

MATERIALS AND METHODS

Bacterial strains and media

P. acnes (ATCC 11827) was used for the screening of antibacterial activity of marine algal extracts. This test strain was cultured with Reinforced Clostridial Medium (Sigma-Aldrich, Tokyo, Japan). Bacteria used for the antibacterial spectrum assay were methicillin-resistant *Staphylococcus aureus* (ATCC 33591), methicillin-sensitive *S. aureus* (ATCC 25923), *Bacillus subtilis* (IFO 14419), *Escherichia coli* (NBRC 12734), *Enterococcus faecium* (NBRC 3826), *Enterococcus faecalis* (NBRC 3971), *Enterococcus serolicida* (NG 8206), *Streptococcus*

mutans (NBRC 13955), *Streptococcus pneumoniae* (GTC 261), *Streptococcus pyogenes* (GTC 262), *Pseudomonas aeruginosa* (IFO 13736) and *Vibrio alginolyticus* (V7) as well as two strains of *P. acnes* (ATCC 1187 and ATCC 25746). These strains were cultured with Tryptic Soy Agar medium (Difco Laboratories, Detroit, MI, USA), except for *P. acnes* and *V. alginolyticus* which was cultured in ZoBell 2216E agar medium.¹⁰

Marine algae collection and preparation of extracts

During the period from April 1994 to August 2003, marine algae samples were collected at 96 points from north to south along the Japanese coastline and were stored at -20 °C until needed. The marine algae extracts were prepared as described in our earlier paper.¹¹

Screening assay for anti-P. acnes activity

Sterilized paper disks (φ 8 mm; Advantec, Tokyo, Japan) permeated with 50 µl each of phosphate-buffered saline or MeOH algal extract and dried completely were placed on double-layer agar plates inoculated with $5.0 \times 10^6 \, \text{CFU} \, \text{ml}^{-1}$ of *P. acnes* strain. These plates were incubated at 37 °C for 2 days under anaerobic condition. After incubation, the zones of inhibition were measured and recorded.

Isolation and purification of Sargafuran

One liter of the MeOH extract from the marine brown alga *Sargassum macrocarpum* (250 g wet weight) was partitioned with a chloroform/water (1:1) mixture. The chloroform/MeOH fraction was concentrated to dryness and redissolved in *n*-hexane/acetone (6:1), and then subjected to a silica gel-60 (Merck, Darmstadt, Germany) column chromatography under continuous elution with *n*-hexane/acetone (6:1, 5:1 and 4:1) and MeOH. The active fractions were pooled and concentrated, and then further chromatographed

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on a second silica gel column, eluting with *n*-hexane/acetyl ethyl ester (10:1, 6:1 and 3:1) and MeOH, and on a reverse-phase Cosmosil (Nacalai Tesque Inc., Kyoto, Japan) column, eluting with 60–80% MeOH containing 0.1% trifluor-oacetic acid (Sigma-Aldrich). The active fractions were finally purified with an HPLC on a reversed-phase column (Mightysil RP-8 GP, \emptyset 4.6×250 mm; Kanto Chemical Co. Inc., Tokyo, Japan), eluting with a gradient of acetonitrile and water containing 0.1% trifluoroacetic acid.

Spectrometric analyses of Sargafuran

Optical rotations were measured with an SEPT-200 polarimeter (Horiba, Tokyo, Japan). UV spectra were recorded on a U-3210 spectrophotometer (Hitachi, Tokyo, Japan) and IR spectra on a model 1720 spectrometer (Perkin-Elmer, Waltham, MA, USA). NMR spectra were recorded in CDCl₃ on a JEOL lambda 500 NMR spectrometer (JEOL, Tokyo, Japan). Chemical shifts are shown as δ values from TMS as the internal reference. Peak multiplicities are quoted in Hz. Mass spectra were measured on a JMS-700 spectrometer (JEOL).

Antibacterial test

The MICs of Sargafuran were determined by the standard microdilution method described by the National Committee for Clinical Laboratory Standards,¹² using a Tryptic Soy Broth medium (Difco Laboratories), except for *P. acnes* and *V. alginolyticus* which was in ZoBell 2216E broth medium. The final volume of Tryptic Soy Broth, Reinforced Clostridial Medium or ZoBell broth medium containing Sargafuran was 100 µl per well to give a starting inoculum density of 5×10^5 CFU ml⁻¹.

Time-kill curve experiment

The time-kill experiment was conducted by the method described by Aeschlimann and Rybak¹³ and Etenza *et al.*¹⁴ The experiments were conducted in test tubes containing 15 ml of fresh Reinforced Clostridial Medium inoculated with an overnight culture of *P. acnes* (ATCC 11827) to give an initial bacterial density of 10⁶ cells per ml. The inoculation was carried out immediately after the addition of Sargafuran or Clindamycin (Sigma-Aldrich) at the final concentrations of MIC, 2×MIC and 4×MIC and incubated at 37 °C for 2 days under anaerobic condition.

Bacteriolytic assay

The seed culture of *P. acnes* (ATCC 11827) on an RCA plate was resuspended in Reinforced Clostridial Medium and washed twice with sterile 10 mM Tris-HCl buffer (pH 7.6). The absorbance was adjusted to 0.1 at 660 nm. The bacterial cell suspension (5 ml each) was aliquoted in sterile test tubes and exposed to Sargafuran at various concentrations or to Achromopeptidase (Wako Pure Chemical Industries Ltd, Tokyo, Japan) at 134 µg ml⁻¹ as a positive control. Untreated bacterial suspensions were used as negative controls. The concentration of MeOH (the solvent of Sargafuran) in each tube was less than 0.1% (v/v). The test tubes were incubated at 37 °C under anaerobic conditions. The absorbance at 660 nm was measured at 0, 0.15, 0.5, 0.45, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 and 12 h after incubation, and the relative absorbance was calculated by dividing each absorbance by that of the negative control. Each treatment was conducted in duplicate.

Cytotoxicity test

The cytotoxic activity of Sargafuran was evaluated as described earlier.¹⁵ Human normal dermal fibroblasts (Morinaga Institute of Biological Science, Yokohama, Japan) were prepared at 5×10^3 cells per 100 µl per well in a 96-well plate. Then Sargafuran, serially diluted twofold, was added to give final concentrations ranging from 3.8 to 480 µg ml⁻¹. The cells were incubated at 37 °C for 3 days. The proliferation of cells was evaluated by an 3-[4,5-dimethylthiazol-2-ly]-2,5-diphenyltetrazolium bromide (MTT) assay¹⁶ and the growth rate relative to the control treatment was calculated.

RESULTS

Marine algae with anti-P. acnes activity

We found 13 species of marine algae with anti-*P. acnes* activity, based on disk diffusion assays, using the extracts from a total of 342 species

of marine algae collected from the Japanese coastline. The algae are *Laurencia brongniartii, Laurencia okamurae, Osonthalia corymbifera, Rhodomela teres, Dictyopteris divaricata, Dictyopteris undulata, Ishige okamurai, Padina crassa, Sargassum fulvellum, S. macrocarpum, Sargassum siliquastrum, Sargassum yezoense* and *Zonaria diesingiana,* showing approximate MIC ranges of 62.5–1000 μ g per disk against *P. acnes* (Table 1). Most of the positive marine algae were brown algae and showed a relatively high anti-*P. acnes* activity at 62.5–250 μ g per disk.

Purification of the antibacterial compound, Sargafuran, from *S. macrocarpum*

As described above, several marine algae showed promising anti-*P. acnes* activity. One of those positive algae, *S. macrocarpum*, is widely distributed throughout the Japanese coastline and can be collected in large amounts during any season. Thus, we selected this algal species to proceed with the isolation and purification of the antibacterial compound against *P. acnes* from the MeOH extract. After several purification steps, starting from 250 g wet weight of *S. macrocarpum*, 3.6 mg of the antibacterial compound, which we named Sargafuran, was obtained.

Structure determination of Sargafuran

Sargafuran was isolated as a colorless oil [UV λ_{max} (CHCl₃) nm (log ε): 241 (4.27), 263 (3.79); [α]_D-1.26°(CHCl₃, *c* 0.95)]. The molecular formula was determined to be C₂₇H₃₆O₄ (found 424.2593, calcd. 424.2614) by high resolution electron impact mass spectrometry (HR-EI-MS). The IR spectrum showed absorption bands at 1686 and 3400 cm⁻¹ attributable to a carbonyl and hydroxyl group, respectively. ¹H- and ¹³C-NMR spectroscopic data indicate that Sargafuran has five singlet methyl groups (C-18, C-19, C-21, C-26 and C-27), seven olefinic bonds, six aliphatic methylenes, a carbonyl group and a quaternary carbon attached to oxygen. Taken together with the presence of seven olefinic bonds and a carbonyl group, 10 units of unsaturation in Sargafuran indicate that the compound has two rings in its structure. ¹H-¹H COSY spectrum revealed the linkages among the methylenes, and olefinic protons (H-6 to H-8, H-10 to H-12 and H-14 to H-16), in addition to two sets of

Table 1 MeOH extracts from marine algae exhibited potent anti-*Propionibacterium acnes* activity

Marine algae	Anti-P. acnes activity ^a
Red algae	+++
Laurencia brongniartii	+++
Laurencia okamurae	++
Odonthalia corymbifera	+++
Rhodomela teres	
Brown algae	
Dictyopteris divaricata	++
Dictyopteris undulata	++
Ishige okamurai	++
Padina crassa	+++
Sargassum fulvellum	++
Sargassum macrocarpum	++
Sargassum siliquastrum	++
Sargassum yezoense	++
Zonaria diesingiana	+

^a+++, \ge 19-mm inhibition zone; ++, 12- to 18-mm inhibition zone; +, \le 11-mm inhibition zone; anti-*P. acnes* activities of the MeOH extracts were evaluated by the disk diffusion method.

connection between olefinic protons (H-2/H-3 and H-23/H-24). The connections in the triene chain (C-6 to C-21) were established from the HMBC correlations (H-8/C-10, 19, H-19/C-9, 10, H-10/C-9, H-11/C-9, 13, H-12/C-14, 20, H-14/C-13, 20, H-18/C-16, 17 and H-21/C-16, 17) among the olefinic methines (H-8, H-12 and H-16), methyl groups (H-18, -19 and -21) and a carbonyl carbon (C-20). The geometry of olefinic bonds was determined to be E (C-8/C-9) and Z (C-12/C-13), respectively, by NOESY correlations (H-7/H-19, H-8/H-10 and H-12/H-14). The 2-methyl furan moiety (C-22 to C-26) was revealed by the HMBC correlations (H-23/C-22, 25, H-24/C-22, 25 and H-26/C-24, 25) and the small $J_{\rm H-H}$ coupling value between H-23 and H-24 (J=3Hz). Additional HMBC correlations (H-2/C-1, -4, H-3/C-1, -4 and H-27/C-1, -2, -5) indicated the presence of 1-methylcyclopenta-2,4-dienol moiety (C-1 to C5, and C-27). The remaining carbonyl carbon (C-20) was attributed to a carboxylic acid. The cyclopentadiene moiety was connected with C-6 of the triene chain at C-5 and with C-22 of methylfuran at C-4 on the basis of HMBC correlations (H-7/C-5, H-6/C-1, H-3/C-22 and H-23/C-4), and this was further supported by a strong fragment ion at m/z 175 (100%) in the EI-MS spectrum. Thus, the structure of Sargafuran was determined as shown in Figure 1. The assignments of ¹H- and ¹³C-NMR signals are indicated in Table 2.

Minimum inhibitory concentration

Sargafuran showed antibacterial activity against *P. acnes* at $15 \,\mu g \,m l^{-1}$. At the same concentration, Sargafuran also showed antibacterial activity against other Gram-positive bacteria, S. pyogenes and S. pneumoniae, and a Gram-negative bacterium, V. alginolyticus (Table 3). However, this compound did not inhibit the bacterial growth of the Gram-negative bacteria, E. coli and P. aeruginosa, and another Gram-positive bacterium, S. mutans, even at a concentration of $120 \,\mu g \, m l^{-1}$.

Bactericidal activity of Sargafuran

The time-kill study showed that Sargafuran decreased the bacterial counts of *P. acnes* at $1/2 \times MIC$ (7.5 µg ml⁻¹) during a 16-h exposure (Figure 2). At MIC $(15 \,\mu g \,m l^{-1})$, Sargafuran completely killed the P. acnes strain tested within 4h. However, Clindamycin did not show bactericidal activity against the P. acnes strain tested even after 72 h of incubation at $4 \times MIC (0.1 \,\mu g \, m l^{-1})$. This bactericidal property of Sargafuran seems to be very promising for developing new types of antibiotics against the multi-drug-resistant P. acnes strains.

Bacteriolytic activity

The reduction in the absorbance of P. acnes cell suspensions was not observed in the presence of Sargafuran, at up to $4 \times MIC$ (60 µg ml⁻¹), until the end of the incubation period (Figure 3). In contrast, the absorbance of P. acnes cell suspensions treated with Achromopeptidase at 134 µg ml⁻¹ as the positive control was reduced drastically in the early incubation period. These results indicated that Sargafuran did not lyse P. acnes cells.

Cvtotoxicity

An MTT assay showed that Sargafuran was not cytotoxic to human normal dermal fibroblast cells (Figure 4). At concentrations up to

Table 2 ¹H- and ¹³C-NMR data of Sargafuran

No.	^{1}H	¹³ C
1	2.13 (3H, <i>s</i>)	15.5 (CH ₃)
2		144.9 (C)
3	6.47 (1H, d, 3.0)	117.1 (CH)
4	6.32 (1H, d, 3.0)	110.3 (CH)
5		148.5 (C)
6		122.9 (C)
7	6.24 (1H, d, 10.0)	121.3 (CH)
8	5.57 (1H, d, 10.0)	130.7 (CH)
9		77.8 (C)
10		126.4 (C)
11	1.67 (2H, <i>m</i>)	40.7 (CH ₂)
12	2.11 (2H, <i>m</i>)	22.6 (CH ₂)
13	5.13 (1H, dd, 6.0, 7.3)	124.9 (CH)
14		134.4 (C)
15	2.06 (2H, <i>t</i> , 7.5)	39.1 (CH ₂)
16	2.58 (2H, <i>m</i>)	28.1 (CH ₂)
17	5.99 (1H, dd, 7.5, 7.0 Hz)	145.7 (CH)
18		130.5 (C)
19	2.26 (2H, t, 7.5)	34.5 (CH ₂)
20	2.11 (2H, <i>m</i>)	27.9 (CH ₂)
21	5.09 (1H, t, 7.0)	123.0 (CH)
22		132.3 (C)
23	1.67 (3H, <i>s</i>)	25.7 (CH ₃)
24	1.35 (3H, <i>s</i>)	25.9 (CH ₃)
25	1.57 (3H, <i>s</i>)	15.8 (CH ₃)
26		173.3 (C)
27	1.58 (3H, <i>s</i>)	17.7 (CH ₃)



Figure 1 2D NMR correlations and mass fragmentation of Sargafuran.

Table 3 Comparative antibacterial activities of Sargafuran and Clindamycin

	MIC (ıgm⊢¹)
Bacterial strain	Sargafuran	Clindamycin
Propionibacterium acnes (ATCC 11827)	15	0.03
P. acnes (ATCC 25746)	15	0.67
MSSA (ATCC 25923)	30	0.47
MRSA (ATCC 33951)	30	>120
Bacillus subtilis (IFO 14419)	30	1.88
Escherichia coli (NBRC 12734)	>120	>120
Enterococcus faecalis (NBRC 3971)	30	>120
Enterococcus faecium (NBRC 3826)	60	0.90
Enterococcus serolicida (NG 8206)	60	1.88
Streptococcus pyogenes (GTC 262)	15	7.50
Streptococcus pneumoniae (GTC 261)	15	7.50
Streptococcus mutans (NBRC 13955)	>120	3.75
Pseudomonas aeruginosa (IFO 13736)	>120	>120
Vibrio alginolyticus (V-7)	15	30



Figure 2 Comparative bactericidal activities of Sargafuran and vancomycin against *P. acnes* (ATCC 11827). **A**, Negative control (antibacterial substance not added); **•**, $1/2 \times MIC$; **•**, $1 \times MIC$; **■**, $4 \times MIC$. The values with the standard error bars are mean values from duplicate experiments.

 $2\times$ the MIC (30 µg ml⁻¹), it did not significantly affect cell viability and did not reduce cell viability by 50% at any concentrations up to 480 µg ml⁻¹.



Figure 3 Bacteriolytic activity of Sargafuran against *P. acnes* (ATCC 11827). •, Positive control $(134 \,\mu\text{g}\,\text{ml}^{-1})$ Achromopeptidase; \blacklozenge , $1 \times \text{MIC}$; \blacksquare , $2 \times \text{MIC}$; \blacktriangle , $4 \times \text{MIC}$. Relative absorbance was calculated by dividing the absorbance of the treated tube by that of the negative control tube. The values with the standard error bars are mean values from triplicate experiments.



Figure 4 Cytotoxicity of Sargafuran to human normal dermal fibroblast cells.

DISCUSSION

Acne vulgaris is a multifactorial disease with an unclear etiology and pathogenesis. The factors known to cause acne vulgaris include follicular hyperkeratosis, sebum secretion, *P. acnes* and inflammation.¹⁷ Recently, cosmetics and toiletries containing natural products, such as extracts of herbs, Chinese plant medicine and seaweed, have been prevalent in commercial markets because of consumer concerns over synthetic chemical ingredients.

This study was conducted to screen marine algae collected from the Japan coastline for antibacterial activity against *P. acnes* to develop new types of cosmetics to treat acne. We found a new anti-*P. acnes* compound, Sargafuran, from the marine brown alga, *S. macrocarpum*. We also have reported two neurostimulating substances, sargaquinoic acid and sargachromenol, from this alga^{18,19} and determined their structures to be 2-methylquinone-type compounds having side chain (C-20 unit). The 2-methylquinone moiety and C-20 side chain of these compounds would be biosynthesized from sikimic acid (C-7 unit) and geranyl geranyl diphospate (C-20 unit). It is likely that the

biosynthetic pathway of Sargafuran carbon skeleton would share with the pathway of sargaquinoic acid and sargachromenol skeletons, because Sargafuran has a similar side chain and the same number of carbons. Sargafuran was stable against heating up to 60 °C, pH 4-7 and irradiation for 24 h. These properties are suitable for cosmetics or skin care products to prevent acne. Clindamycin is generally used for curing acne as a clinical treatment. However, Clindamycin was shown to be bacteriastatic even at 4×MIC, whereas Sargafuran showed bactericidal activity against P. acnes. This is a superior property because bactericidal activity minimizes the chance of development of resistance. Takahashi et al.²⁰ have also reported that eucalyptus leaf extracts and constituent flavonoids showed great antibacterial activity against Gram-positive bacteria, including P. acnes. The marine brown alga, S. macrocarpum, containing Sargafuran might be a good candidate from which to develop and produce the anti-acne cosmetics or skin care products in the near future.

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Discovery and antibacterial activity of glabramycin A–C from *Neosartorya glabra* by an antisense strategy

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Treatment of drug-resistant bacteria is a significant unmet medical need. This challenge can be met only by the discovery and development of new antibiotics. Antisense technology is one of the newest discovery tools that provides enhanced sensitivity for detection of antibacterials, and has led to the discovery of a number of interesting new antibacterial natural products. Continued utilization of this technology led to the discovery of three new bicyclic lactones, glabramycins A–C, from a *Neosartorya glabra* strain. Glabramycin C showed strong antibiotic activity against *Streptococcus pneumoniae* (MIC 2 μ g ml⁻¹) and modest antibiotic activity, and plausible biogenesis of these compounds have been discussed.

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INTRODUCTION

The discovery and development of clinically useful antibiotic classes, such as, the aminoglycosides, macrolides and tetracyclines, have clearly shown that bacterial protein synthesis is a viable target for antibacterial drug discovery.^{1,2} The bacterial ribosome responsible for protein synthesis consists of a small 30S and a large 50S subunit. The ribosomal subunit (aka S4), referred to as RpsD, is a component of the 30S subunit.^{3–5} The RpsD protein is encoded by the *rpsD* gene, which is essential and resides in an operon containing only one other gene, *SAV1718*, which is a nonessential gene.^{6,7} Therefore, this gene was selected as a drug-target for the discovery of antibacterial agents.

For antibiotic discovery targeting the rpsD gene, we constructed a Staphylococcus aureus S1-782B strain expressing antisense RNA under xylose control, leading to hypersensitivity against RpsD inhibitors. To implement this approach for drug discovery, we designed a two-plate assay in which one plate was seeded with an rpsD antisense S. aureus strain and the other with an S. aureus EP 167 (control) strain. A similar general methodology led to the discovery of platensimycin and platencin.⁸⁻¹³ The screening of over 138 000 microbial extracts against rpsD two-plate antisense whole-cell assay led to the isolation of a series of interesting new compounds exemplified by lucensimycins,14-16 coniothyrione,¹⁷ pleosporone,¹⁸ phaeosphenone,¹⁹ and okilactomycin.²⁰ Continued screening and follow-up of one of the active extracts produced by Neosartorya glabra led to the isolation of three macrolactones namely, glabramycin A-C (1-3, Figure 1). The isolation, structure elucidation, relative configuration and antibacterial activity of the glabramycins (1-3) are described.

RESULTS AND DISCUSSION

The producing organism was isolated from a soil sample collected from Candamia, Spain. The strain was identified as *Neosartorya glabra* by sequencing the large subunit of DNA of the D1D2 region and by tracing the phylogenetic relationships. The strain was grown in a submerged fermentation for 14 days and compounds were extracted with an equal volume of acetone. The extract was chromatographed on Amberchrome, a capture resin, followed by reversed-phase C₈ HPLC to give **1** (7 mg, 7 mgl⁻¹), **2** (1.9 mg, 1.9 mgl⁻¹) and **3** (1 mg, 1 mgl⁻¹), each as yellow gums.

Glabramycin A (1), the most abundant of the three compounds, showed a molecular formula $C_{22}H_{24}O_6$ as deduced from the HRESIFTMS data. The ¹³C NMR spectrum and distortionless enhancement by polarization transfer (DEPT) analysis of 1 showed signals for one methyl, four methylenes, ten methines and seven quaternary carbons (Table 1). The ¹H NMR spectrum showed signals for four olefinic methine doublet of doublets with J=15 and 11 Hz, and two methine doublets with I=15 Hz. The COSY correlations of these six olefinic methines indicated the presence of an E-triene in which each end terminated with a quaternary carbon. The methine $(\delta_H 5.90)$ at one end of the triene chain showed heteronuclear multiple bond correlations (HMBCs) to a carboxyl carbon at $\delta_{\rm C}$ 170.3, and a carboxylic group was thus placed at one end of the triene. The terminal methine ($\delta_{\rm H}$ 6.00) at the other end of the triene chain showed HMBC correlations to two quaternary olefinic carbons resonating at δ_C 167.4 and δ_C 108.6, allowing for further extension of the triene chain.

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Figure 1 Structures and relative configurations of glabramycins A-C (1-3).

Table 1	¹ H (600 MHz) and	¹³ C (125 MHz)-NMR	assignments of	f glabramycin	A (1),	B (2) a	nd C	(3)
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	1			2		3
	1	CD ₃ OD	2	CD ₃ OD	3	CDCI ₃
Position	δ_{C}	δ _H , mult, <i>J</i> in Hz	δ_{C}	δ _{H,} mult, <i>J</i> in Hz	δ_{C}	δ _{H,} mult, <i>J</i> in Hz
1	170.3		170.1		170.8	
2	124.4	5.90, d, 15	124.5	6.00, d, 15	122.2	5.90, d, 15
3	145.6	7.30, dd, 15, 11	145.2	7.30, dd, 15, 11	145.5	7.40, dd, 15, 11
4	136.0	6.60, dd, 15, 11	136.2	6.60, dd, 15, 11	134.5	6.57, dd, 15, 11
5	140.4	6.54, dd, 15, 11	140.3	6.80, dd, 15, 11	140.0	6.75, dd, 15, 11
6	136.9	7.10, dd, 15, 11	137.3	7.00, dd, 15, 11	135.9	6.97, dd, 15, 11
7	129.1	6.0, d, 15	123.3	7.20, d, 15	122.9	7.32, d, 15
8	167.4		164.5		163.1	
9	108.6		110.9		112.0	
10	125.3		49.4	3.46, t, 9.6	44.0	3.62, dd, 10, 4.5
11	124.9	6.49, dd, 7.5, 1.0	82.0	5.00, m	77.5	5.09, dd, 8.5, 4.5
12	126.9	6.95, t, 7.5	35.9	2.20, m, 2.40, m	137.6	6.87, dd, 10.5, 4.5
13	115.9	6.73, dd, 10, 7.5	26.0	2.20, m, 2.40, m	132.5	6.23, d, 10.5
14	157.4		213.3		198.7	
15	131.8		50.6	2.60, m	48.3	2.46, m
		2.07, m		1.46, m,		2.05, m
16	27.3	2.70, m	23.7	1.62, m	22.9	1.40, m
		1.94, m		1.80, m		1.53, m
17	29.5	1.87, m	25.2	1.30, m	23.6	1.89, m
		1.34, m		1.78, m		1.65, m
18	22.6	1.53, m	24.9	1.88, m	32.4	1.76, m
		1.49, m		1.56, m		1.38, m
19	33.9	1.83, m	32.9	1.83, m	24.8	1.77, m
20	73.5	5.30, m	74.1	4.94, m	73.2	5.00, m
22	164.7		170.1		164.9	
23	18.5	1.10, d, 6.5	20.3	1.28, d, 6.5	20.6	1.30, d, 6.5

Three of the remaining olefinic methines coupled with each other, with J=7.5 Hz indicating the presence of a 1, 2, 3-tri-substituted phenyl group. The only non-olefinic methine proton resonating at $\delta_{\rm H}$ 5.30 showed heteronuclear multiple quantum coherence (HMQC) correlation to an oxygenated carbon resonating at $\delta_{\rm C}$ 73.5. The methyl group resonating at $\delta_{\rm H}$ 1.10 showed HMBC correlations to the methine carbon C-20 ($\delta_{\rm C}$ 73.5) and to the methylene at C-19 ($\delta_{\rm C}$ 33.9). This observation allowed for the placement of the methyl group on the oxygenated carbon C-20. The COSY correlations from the methyl protons, H₃-23 at $\delta_{\rm H}$ 1.10 to the methylene H₂-16 at $\delta_{\rm H}$ 2.07 and $\delta_{\rm H}$ 2.7, established the C16-C20-C23 spin system. The HMBC correlations of the terminal methylene protons of the alkyl chain at $\delta_{\rm H}$ 2.70 and $\delta_{\rm H}$ 2.07 to carbons $\delta_{\rm C}$ 131.8 (C-15), 125.3 (C-10) and 157.4

(C-14) established its connection to the phenyl group. The HMBC correlation of the methine proton ($\delta_{\rm H}$ 5.30)to an ester carbonyl at $\delta_{\rm C}$ 164.7 (C-22) allowed for the linkage of this carbonyl and the methine in the form of a lactone ring. Finally, the ester carbonyl C-22 was connected to the olefinic carbon C-9 ($\delta_{\rm C}$ 108.6) to form the macrocyclic lactone, to satisfy the degrees of unsaturation and the molecular formula. This assignment was supported by the ¹³C chemical shifts of C-8, C-9 and C-22. The substitution around the phenyl ring was confirmed by HMBC correlations of the aromatic protons and supported by HMBC correlations of the benzylic methylene protons (Figure 2).

The structures for glabramycins, B (2) and C (3), were determined by comparison of ${}^{13}C$ and ${}^{1}H$ NMR spectral data with 1 (Table 1).

Glabramycin B (2) exhibited a molecular formula, $C_{22}H_{26}O_6$. The ¹H NMR spectrum of 2 showed only six olefinic protons of trienoic acid moiety, indicating the presence of a saturated six-membered ring system. The COSY spectrum showed an extended spin system comprising C20(C23)-C15-C10-C14. The HMBC correlations of H-10 (δ_H 3.46), H₂-12 (δ_H 2.20, 2.40) and H₂-16 (δ_H 1.46, 1.62) to the downfield carbonyl C-14 (δ_C 213.3) allowed the placement of a carbonyl group at C-14 in the cyclohexanone ring. The ether bridge between C8 and C11 made the dihydrofuran ring that satisfied the molecular formula. The relative configurations at C-10, C-11 and C-15 were established from the magnitude of the scalar couplings. H-10 resonated as a triplet with *J*=9.6 Hz because of axial–axial couplings both from H-11 and H-15, thus establishing an anti relationship between these three protons and a *trans*-ring fusion.

Glabramycin C (**3**) showed a molecular formula that was isomeric to that of **1**. Comparison of the ¹H and ¹³C NMR spectra of **3** with those of compound **2** indicated the presence of a pair of olefinic protons, $\delta_{\rm H}$ 6.23 (d, *J*=10.5 Hz) and 6.87 (dd, *J*=10.5, 4.5 Hz), which were assigned to H-13 and H-12 by COSY correlations and confirmed by their HMBC correlations to an upfield shifted C-14 carbonyl ($\delta_{\rm C}$ 198.7), thus assigning structure **3** for glabramycin C (Table 1).

On the basis of these data, structures 1, 2 and 3, with relative configurations, were assigned for glabramycins A, B and C, respectively. Biogenetically, these compounds are likely to originate from a



Figure 2 Heteronuclear multiple bond correlations (HMBC) of glabramycin A (1).

Table 2 Antibacterial activities (MIC, µg ml⁻¹)^a of glabramycins A–C (1–3)

Strains ^b	Phenotype	Strain #	1 °	2 °	3 °
S. aureus	meth ^S	ATCC 29213	>64	>64	16
S. aureus	meth ^S	MB2865	>64 (64)	>64	16 (8)
S. aureus (+50% human serum)	meth ^S	MB2865	>64	>64	>32
S. pneumoniae ^d	pen ^s , quin ^s , mac ^s	CL2883	>64	4	2
S. pneumoniae ^e	pen ^s , quin ^s , mac ^s	CL2883	32	64	2
E. faecalis	van ^s , mac ^R	CL8516	>64	>64	>32 (32)
B. subtilis	Wt	MB964	>64	>64	16
H. influenzae	Amp ^s , quin ^s , mac ^s	MB4572	>64	>64	>64
E. coli	Wt	MB2884	>64	>64	>32
E. coli envA/tolC	Wt	MB2884	64	>64	>32
C. albicans	Wt	MY1055	>64	>64	>32

Abbreviations: CAMHB, cation adjusted Mueller-Hinton broth; MIC, minimum inhibitory concentration; MIC80, minimum inhibitory concentration that inhibits 80% growth

^aMIC determined using NLSI protocols.

^bAll strains were tested in CAMHB medium, unless mentioned otherwise, under National Committee for Clinical laboratory Standards (NCCLS) guidelines

^cThe data in parentheses are MIC80 values.

^dCAMHB +2.5% lysed horse blood medium elsosensitet medium polyketide pathway by condensation of 11 acetate units to a undecaketide, which likely undergoes cyclization, reduction and dehydration to produce compound 1, which then produces compounds 3 and 2.

Biological activities

All three glabramycins were tested in the S. aureus antisense rpsDsensitized two-plate differential sensitivity assay. Glabramycin C showed the most potent activity in this assay and showed a minimum detection concentration (MDC) of $62 \mu g m l^{-1}$. At this concentration, a more than 5 mm zone differential was observed with a 10.78 mm zone of inhibition on the antisense plate vs a 5.63 mm zone of inhibition on the control plate. The size of the zone of clearance was dose dependent, and at 500 μ g ml⁻¹, it produced a zone of clearance of 16.89 and 6.8 mm on the antisense and the control plate, respectively. The other two compounds were less active. Glabramycin A was approximately four-fold less active and showed intermediate activity with MDC of 250 µg ml⁻¹, producing zone sizes of 11.24 and 6.25 mm on the antisense and the control plate, respectively. The MDC of glabramycin B was greater than 500 µg ml⁻¹. Glabramycin C showed better activity against a panel of bacteria used in this assay. It inhibited S. aureus growth with MIC values of 16 µg ml⁻¹ (Table 2). Glabramycin C exhibited a similar activity against Bacillus subtilis, but was less active against Enterococcus faecalis (MIC > $32 \,\mu g \, m l^{-1}$). The best activity was against Streptococcus pneumoniae regardless of the medium used, and inhibited the growth with an MIC value of 2 µg ml⁻¹. Glabramycins A and B were significantly less active, see Table 2. None of these compounds inhibited growth of Gram-negative bacteria or fungi Candida albicans. Mechanistically, glabramycin A, the most abundant of the three, showed 2-3-fold preferential inhibition of RNA synthesis (IC₅₀ 10 µg ml⁻¹) compared with DNA and protein synthesis in a macromolecular synthesis assay (Figure 3). Inhibition of RNA synthesis, IC₅₀ $(10 \,\mu g \,m l^{-1})$, of S. aureus is 10 times more potent than the MIC value $(100 \,\mu g \,m l^{-1})$ against the same *S. aureus* strain. Why the compounds discovered in this assay preferentially inhibit RNA synthesis rather than the expected protein synthesis is not clear and requires further investigation.

No compounds have earlier been reported from *N. glabra* strain, but a large number of biologically active compounds have been reported from genera *Neosartorya*, for example, the angiogenesis inhibitor, azaspirene.²¹ It seems that the most chemically studied



Figure 3 Macromolecular synthesis inhibition by glabramycin A (1) in *S. aureus.* No significant preference to inhibition of DNA, RNA and protein synthesis was observed.

species is *Neosartorya fischeri*, which is known to produce substance P inhibitors namely the fiscalins,²² the toxin fischerin²³ and the tremorgenic mycotoxins, fumitremorgins,²³ neosartorrin,²⁴ and terreins.²⁵

In conclusion, it is evident that the antisense screening approach provides higher sensitivity and allows the discovery of antibiotics with weaker activity, exemplified by the discovery of the three new compounds, glabramycins A–C. One of the three compounds showed moderate antibacterial activity without being a general toxin and can be exploited further. Screening using an antisense sensitized *rpsD* strains of *S. aureus* led to the isolation of a number of antibacterial agents that do not preferentially inhibit protein synthesis but often RNA synthesis. Understanding of this phenomenon requires further study.^{14,15,17–19} Although lack of selectivity for protein synthesis inhibition by compounds discovered by *rpsD*-sensitized antisense screening is perplexing, it cannot be merely explained as a generalized technology artifact, because a similar screening approach using a *fabF*-sensitized strain led to the discoveries of platensimycin and platencin, highly selective fatty-acid synthesis inhibitors.^{10,11}

EXPERIMENTAL SECTION

For general experimental procedure, see, for example, Zhang et al.²⁶

Producing organism

The *N. glabra* (MF7030, F-155,700) strains were isolated from hot waterpasteurized soil collected from Candamia, near Valdefresno province of León, Spain. The ascomata, ascospores and conidial states were observed on malt yeast extract agar and were readily recognized as a species of *Neosartorya* (Ascomycota, Eurotiales).

The DNA of strain F-155,700 was extracted and used as a template for polymerase chain reaction (PCR) reactions. The D1D2 region of the large subunit of ribosomal DNA (LSU rDNA) was amplified and sequenced to aid in identification and to infer phylogenetic relationships of the strains to other fungi. The sequences were used to query GenBank for similar ribosomal sequences. The best matches with the large subunit (LSU) region and the percentage similarities were: *N. glabra* (U28456) 99%.

Inoculum was prepared by inoculating agar plugs into a 250 ml Erlenmeyer flask containing 60 ml seed medium of the following composition: gl^{-1} in distilled water (corn steep powder, 2.5; tomato paste, 40.0; oat flour; 10.0; glucose, 10.0; FeSO₄·7H₂O, 0.01; MnSO₄·4H₂O, 0.01; CuCl₂·2H₂O, 0.0025; CaCl₂·2H₂O, 0.001; H₃BO₃, 0.00056; (NH₄)₆MoO₂₄·4H₂O, 0.00019; ZnSO₄·7H₂O, 0.01). The pH was adjusted to 6.8 before autoclaving. The seed culture was incubated for 5 days at 22 °C on a gyratory shaker (220 rev min⁻¹)

before the inoculation of the production medium. The production medium, designated asWS80, consisted of g/l in distilled water (whole wheat flour (Pillsbury), 50; xylose, 40; and fructose, 40). The 100 ml medium aliquots was dispensed in 500 ml Erlenmeyer flasks, inoculated with 1% volume of the seed culture, and were agitated for 14 days at 22 °C.

Extraction and isolation of glabramycins

One l fermentation broth (10 flasks) was extracted with one l acetone by shaking for 1 h after the harvest. The acetone extract (21) was evaporated under reduced pressure to less than one liter and further diluted with 500 ml water and loaded onto a 50 cc Amberchrome CG161 m (Rohm & Haas, Reading, PA, USA) column. The column was eluted with a 100 min 10–100% aqueous methanol-gradient. The activity was detected in a late eluting band. The active fraction (600 mg) was further purified by repeated reversed-phase HPLC (50% aqueous CH₃CN + 0.1% TFA on a Zorbax SB C₈, 24×250 mm, Agilent Technologies, Santa Clara, CA, USA). The identical fractions were pooled from 10 runs and lyophilized to afford glabramycin A (7 mg, 7 mg/l), glabramycin B (1.9 mg, 1.9 mg/l) and glabramycin C (1 mg, 1 mg/l) as yellow gums.

Glabramycin A (1): $[\alpha]^{23}_{D}$ +7.7 (*c* 0.44, CH₃OH), UV (CH₃OH) λ_{max} 278 (log ε 3.73), 333 (3.70), IR (ZnSe) ν_{max} 3369, 2936, 1684, 1627, 1580, 1498, 1457, 1377, 1284, 1252, 1210, 1142, 1052, 888 cm⁻¹, High Resolution Electrospray Ionization Fourier Transformation mass spectrometry (HRESIFTMS) (*m/z*) 385.1648 (calcd for C₂₂H₂₄O₆+H, 385.1651), for ¹H and ¹³C NMR see Table 1.

Glabramycin B (2): $[\alpha]^{23}_{D}$ -0.8 (*c* 0.6, CH₃OH), UV (CH₃OH) λ_{max} 269 (log ϵ 3.39), IR (ZnSe) ν_{max} 3445, 2966, 2933, 1722, 1693, 1661, 1465, 1375, 1239, 1190, 1128, 1013 cm⁻¹, HRESIFTMS (*m/z*) 387.1802 (calcd for C₂₂H₂₆O₆+H, 387.1808), for ¹H and ¹³C NMR see Table 1.

Glabramycin C (3): $[\alpha]^{23}_{D}$ –1.3 (*c* 0.6, CH₃OH), UV (CH₃OH) λ_{max} 277 (log ϵ 3.49), IR (ZnSe) ν_{max} 3445, 2965, 2933, 1735, 1657 1466, 1386, 1349, 1307, 1284, 1259, 1184, 1151, 1130, 1094, 1011 cm⁻¹, HRESIFTMS (*m/z*) 385.1647 (calcd for C₂₂H₂₂O₆+H, 385.1651), for ¹H and ¹³C NMR see Table 1.

Two-plate differential sensitivity rpsD assay

Staphyloccus aureus cells (RN450) carrying plasmid S1-782B bearing antisense to *rpsD* (AS-RNA strain) or a vector (control strain) were inoculated from a frozen vial source into a tube containing 3 ml of Miller's LB Broth (Invitrogen, Carlsbad, CA, USA) plus $34 \,\mu g \, ml^{-1}$ of chloramphenicol. Tubes were incubated at 37 °C at 220 r.p.m. for 18–20 h and kept at room temperature (23 °C) until use. Miller's LB broth was supplemented with 1.2% Select agar (Invitrogen), 0.2% glucose, 15 $\mu g \, ml^{-1}$ chloramphenicol and 12 mM of xylose (only for the antisense strain). The OD₆₀₀ of the culture was measured and diluted to 1/1000, and an OD 3.0 culture was inoculated. Next, 100 ml of the culture media was poured into each NUNC plate, the well-caster templates were placed into the agar and the agar was allowed to solidify. Thereafter, 20 μ l of the test samples were added to the wells and the plates were incubated at 37 °C for 18 h and zones of inhibition were measured. MDC (minimum detection concentration) values were determined by two-fold serial dilution.

Antibiotic assay (MIC)

The MIC (minimum inhibitory concentration) against each of the strains was determined as described earlier and under the guidelines of the National Laboratory Standards Institute (NLSI). The Cells were inoculated at 10^5 cfu ml⁻¹, followed by incubation at 37 °C with a 2-fold serial dilution of compounds in the growth medium for 20 h. MIC is defined as the lowest concentration of an antibiotic inhibiting visible growth.

Macromolecular synthesis inhibition

The assay was carried out as described earlier. Briefly, mid-log (A_{600} =0.5–0.6) S. *aureus* growths were incubated with an increasing concentration of each inhibitor at 37 °C for 20 min with 1 µCi ml⁻¹ 6-[³H]thymidine, 1 µCi ml⁻¹ 5,6-[³H]uracil or 5 µCi ml⁻¹ 4,5-[³H]leucine, to measure DNA, RNA and protein synthesis, respectively. The reaction was stopped by the addition of 10% trichloroacetic acid and the cells were harvested using a glass fiber filter (Perkin-Elmer Life Sciences, Waltham, MA, USA, 1205-401). The filter was dried and counted with a scintillation fluid.

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Prodigiosin biosynthesis gene cluster in the roseophilin producer *Streptomyces griseoviridis*

Takashi Kawasaki, Fumi Sakurai, Shun-ya Nagatsuka and Yoichi Hayakawa

Streptomyces griseoviridis 2464-S5 produces prodigiosin R1, a tripyrrole antibiotic, and roseophilin, a structurally related compound containing two pyrrole and one furan rings. A gene cluster for the biosynthesis of a prodigiosin was identified in *S. griseoviridis*. The cluster consisted of 24 open reading frames, including 21 genes (*rphD*-*rphZ*) homologous to prodigiosin biosynthesis genes in the *red* cluster in *Streptomyces coelicolor* A3(2). The expression of *rphN* in *S. coelicolor* lacking *redN* restored the production of prodigiosin.

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Keywords: biosynthesis gene; prodigiosin; roseophilin; Streptomyces griseoviridis

INTRODUCTION

Several tripyrrole antibiotics are known to belong to the prodigiosin family.¹⁻⁴ S. griseoviridis 2464-S5 produces prodigiosin R1⁵ and roseophilin,⁶ a structurally related compound containing two pyrrole and one furan rings, as shown in Figure 1. Roseophilin is considered to be biosynthesized partially with the same pathway as prodigiosin R1. Prodigiosin-family biosynthesis genes have been reported to exist in the pig gene clusters in Serratia marcescens ATCC 274 and Serratia sp. ATCC 39006 (which make prodigiosin itself),⁷ in the red gene cluster in S. coelicolor A3(2) (which makes undecylprodiginine)⁸ and in the hap gene cluster in Hahella chejuensis KCTC 2396 (which makes prodigiosin itself).9 PCR amplification of S. griseoviridis genomic DNA has identified genes homologous to redH, redM and redW, involved in the biosynthesis of undecylprodiginine in S. coelicolor (Figure 2).⁵ We report herein the cloning and characterization of the rph gene cluster for prodigiosin biosynthesis from S. griseoviridis 2464-S5.

RESULTS

Amplification of prodigiosin biosynthesis genes from *S. griseoviridis*

In the biosynthesis of prodigiosin, the *redN/pigH* gene product is involved in the formation of the central pyrrole moiety (Figure 2). As roseophilin contains a furan ring in place of the central pyrrole ring in prodigiosins, *redN/pigH* might be characteristic of the prodigiosin biosynthesis. A pair of primers was designed from the conserved amino acid (aa) sequences of *redN/pigH* in the three prodigiosin producers, *S. coelicolor* A3(2), *Serratia marcescens* ATCC 274 and *Serratia* sp. ATCC 39006.^{7,8} PCR amplification of *S. griseoviridis* genomic DNA gave a *redN/pigH*-like gene fragment. Nucleotide sequencing and homology search showed that the fragment shared amino acid identity of 84% with RedN. The prodigiosin biosynthesis genes, *redM* and *redW*, encode L-prolyl-AMP ligase and L-prolyl-PCP dehydrogenase, respectively (Figure 2). These two enzymes have been reported to be commonly involved in the biosynthesis of pyrrole-containing antibiotics.¹⁰ PCR was carried out using several primers designed from the conserved amino acid sequences of L-prolyl-AMP ligase and L-prolyl-PCP dehydrogenase genes in the coumermycin A₁ producer, *Streptomyces rishiriensis* DSM 40489 (*proB* and *proA*),¹¹ in the pyrrolomycin producers, *Streptomyces vitaminophilum* ATCC 31673 (*pyr8* and *pyr7*) and *Streptomyces* sp. Strain UC 11065 (*dox8* and *dox7*)¹² and in the prodigiosin producer, *S. coelicolor* A3(2) (*redM* and *red W*). DNA sequencing of the PCR products identified single genes encoding L-prolyl-AMP ligase and L-prolyl-PCP dehydrogenase in *S. griseoviridis*.

Southern hybridization of genomic DNA with prodigiosin biosynthesis genes

Southern blot analysis was carried out on *S. griseoviridis* genomic DNA to identify *redH-*, *redM-*, *redN-* and *redW-*like genes. Digested genomic DNA with *Bam*HI in agarose gel was hybridized with the PCR products as probes. Each single band was detected by Southern hybridization as shown in Figure 3, suggesting that the strain contained a single set of genes for prodigiosin/roseophilin biosynthesis.

Cloning of a prodigiosin biosynthesis gene cluster from *S. griseoviridis*

A cosmid clone was selected by colony hybridization and Southern blot analysis from a cosmid library using the *redW*-like gene fragment as a probe. Nucleotide sequencing identified genes homologous to *redY*, *redX* and *redD*, as well as *redW*, in a *SacI* 6.0-kbp fragment of the cosmid (Figure 4). Further Southern hybridization was carried out on

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Prodigiosin



Roseophilin

Prodigiosin R1

Ń=

H₃C

CH₃





Undecylprodiginine

Butyl-meta-cycloheptylprodiginine

Figure 1 Structures of roseophilin and prodigiosins.



Figure 2 Proposed pathway of prodigiosin biosynthesis.

various cosmid fragments using the redH-, redM- and redN-like gene fragments as probes. Selected cosmid fragments were sequenced to identify a gene cluster containing 24 open reading frames (ORFs) as shown in Figure 4. Homology search showed that 21 of the 24 ORFs were homologous to the red genes involved in the biosynthesis of undecylprodiginine and butyl-meta-cycloheptylprodiginine in



Figure 3 Southern blot analysis of S. griseoviridis genomic DNA.



Figure 4 Prodigiosin biosynthesis gene cluster (rph cluster) from S. griseoviridis.

S. coelicolor A3(2),⁸ and the corresponding genes in S. griseoviridis were designated as rphD-rphZ (Table 1). In the cluster, sequence homology of rphA1 and rphA2 to ABC-transporter genes assigned them to self-resistance genes. A regulatory gene, rphA3, was also identified from sequence homology to a transcriptional regulator gene in *Mycobacterium ulcerans* (Table 1).

Expression of rphN in S. coelicolor lacking redN

To establish the function of the *rph* genes, we disrupted the *redN* gene in *S. coelicolor* A3(2). The disruption plasmid, pRedN-DIS, was constructed (Figure 5) and introduced into *S. coelicolor*. Three colorless colonies were selected on R2YE plates and the *redN* gene disruption was confirmed by PCR analysis using their genomic DNAs as templates.

The *rphN* expression vector, pWHM-rphN, was constructed and introduced into *S. coelicolor* lacking *redN* (*S. coelicolor* Δ *redN*). *S. coelicolor* Δ *redN*, harboring pWHM-rphN or an empty vector (pWHM860), and the wild-type strain were cultivated and the mycelial extracts were analyzed by HPLC. The *redN* disruptant expressing *rphN* produced undecylprodiginine, a major prodigiosin from the wild type (Figure 6), thereby showing that *rphN* can act as a prodigiosin biosynthesis gene in *S. coelicolor*.

DISCUSSION

The rph gene cluster consisted of 21 genes homologous to prodigiosin biosynthesis genes in S. coelicolor, two self-resistance genes and a regulatory gene. The putative functions and locations of orf1 (DNA helicase) and orf3 (RNA methyltransferase) suggested that orf1, orf2, orf3, orf4 and orf5 are excluded from the biosynthesis gene cluster (Table 1 and Figure 4). The rph cluster did not contain genes homologous to redC, redE, redF, redQ, redR or redS in S. coelicolor (Figure 7). Among them, redQ and redR have been reported to be involved in the biosynthesis of an alkyl side chain and not to be essential for the biosynthesis of undecylprodiginine.¹³ Although the redF-homologous gene, pigN, was involved in O-methylation, two pigN mutants reduced, but retained, prodigiosin productivity.¹⁴ No role has been assigned to redS in the prodigiosin pathway and only the 146 aa N-terminal region of PigB (671 aa) shows sequence similarity with RedS.7 In Serratia sp., PigB catalyzes the formation of 2-methyl-3-n-amylpyrrole (MAP), which is replaced by 2-undecylpyrrole in S. coelicolor.14 No role has been assigned to redC or to redE, and no homologous genes have been found in the pig cluster in Serratia sp.7 These findings suggest that these genes are nonessential for the biosynthesis of prodigiosins, and the rph gene cluster identified here contains all the genes for prodigiosin biosynthesis in

Table 1 Deduced functions of gene products in the rph cluster

Gene products	Homologous protein	Identity (%)
Orf1	ATP-dependent DNA helicase of Streptomyces avermitilis	77
Orf2	Uncharacterized protein of S. avermitilis	47
RphY	Hypothetical protein (RedY)	55
RphW	L-Prolyl-PCP dehydrogenase (RedW)	69
RphX	β-Ketomyristoyl-ACP synthase (RedX)	47
RphD	Transcriptional regulator (RedD)	61
RphA1	ABC-transporter of Mycobacterium ulcerans	46
RphA2	ABC-transporter of M. ulcerans	57
RphA3	Transcriptional regulator of M. ulcerans	40
RphM	L-Prolyl-AMP ligase (RedM)	54
RphL	Polyketide synthase (RedL)	54
RphK	Oxidoreductase (RedK)	68
RphJ	Thioesterase (RedJ)	58
RphG2	Oxygenase (RedG)	51
RphG3	Oxygenase (RedG)	39
RphG4	Oxygenase (RedG)	44
RphI	O-methyltransferase (RedI)	59
RphH	Pyrrole-condensing enzyme (RedH)	59
RphG	Oxygenase (RedG)	57
RphZ	Response regulator (RedZ)	50
RphV	Dehydrogenase (RedV)	42
RphU	Phosphopantetheinyl transferase (RedU)	46
RphT	Hypothetical protein (RedT)	63
RphP	3-Oxoacyl-ACP synthase (RedP)	58
RphO	Peptidyl carrier protein (RedO)	75
RphN	4-Hydroxy-2,2'-bipyrrole-4-methanol synthase (RedN)	78
Orf3	RNA methyltransferase of Brucella abortus	31
Orf4	Uncharacterized protein of Streptomyces coelicolor	54
Orf5	Integral membrane protein of S. coelicolor	63

S. griseoviridis. Moreover, primers designed from redS/PigB and redF/ pigN did not amplify S. griseoviridis genomic DNA (data not shown).

The rph cluster included four putative oxygenase genes (rphG, rphG2, rphG3 and rphG4) homologous to redG, which was assumed to catalyze the cyclization of an alkyl side chain for the biosynthesis of butyl-meta-cycloheptylprodiginine.8 The same function is required by the high-sequence similarity of rphG and redG and their locations (Table 1 and Figure 7). The difference in the number and position of carbon-carbon bridges between prodigiosin R1 and roseophilin might be explained by the presence of multiple oxygenase genes.

The rph cluster did not contain biosynthesis genes characteristic of roseophilin. Such genes are involved in the furan ring formation and chlorination. The PCR and Southern hybridization analyses, however, detected single genes homologous to redH, redM and redW, which are considered to be common to both prodigiosin and roseophilin biosynthesis. These results suggested the existence of another gene cluster to complement the biosynthesis of roseophilin. Search for the second gene cluster for the roseophilin biosynthesis is in progress.

METHODS

Bacterial strains, plasmids and culture conditions

S. griseoviridis 2464-S5 and S. coelicolor A3(2) were used as sources of total DNA for the cloning experiment. Media and growth conditions for S. griseoviridis were described earlier.⁶ Prodigiosin production by S. coelicolor was assessed on R2YE plates.¹⁵ Yeast extract malt extract (YEME) liquid culture was used for genomic DNA isolation and for making protoplasts from S. coelicolor.¹⁵ Escherichia coli XL1-blue MRF' and the plasmids, pGEM-5Z,



Figure 5 Disruption of the redN gene in S. coelicolor. Molecular marker (lane 1), PCR product from the wild-type strain (lane 2) and PCR products from the redN disruptants (lane 3-5). Amplified fragments are shown schematically. Arrows indicate primers used for PCR analysis. B: Bg/II site. X: Xbal site. H: HindIII site.





Figure 6 HPLC analysis of prodigiosins produced by the redN disruptant expressing rphN.

pGEM-7Z, pGEM-11Z, pUC118 and pUC119 were used for sequencing analyses.

DNA isolation and manipulation

All restriction enzymes, T4 DNA ligase and calf intestinal alkaline phosphatase were obtained from Nippon Gene (Toyama, Japan). Transformation of E. coli

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with plasmid DNA by electroporation was carried out under standard conditions using a Gene Pulser II electroporation system (Bio-Rad, Hercules, CA, USA). Other general procedures were carried out as described by Sambrook *et al.*¹⁶

Amplification and cloning of prodigiosin biosynthesis genes

A *redN*-like gene fragment was amplified using *S. griseoviridis* genomic DNA and a pair of primers (5'-GAYGGSGTSTTYWSSATGCAYGG-3' and 5'-SGGRWASACSACSGTYTGRCA-3'). The reaction mixture consisted of 1× GC buffer I (TaKaRa, Kyoto, Japan), 0.4 mM dNTP (deoxyribonucleotide triphosphate) mixture, 2 μ M of each primer, 0.5 ng μ l⁻¹ genomic DNA and 0.05 U μ l⁻¹ *LA Taq* DNA polymerase (TaKaRa). After heating at 95 °C for 4 min, PCR was carried out for 30 cycles (95 °C 1 min, 50 °C 30 s, 72 °C 45 s). After purification with QIAquick Gel Extraction Kit (Qiagen, Hilden, Germany), the PCR products were linked with pGEM-T Easy Vector (Promega, Madison, WI, USA) using a T4 DNA ligase. *E. coli* was transformed with the plasmid DNA and selected clones were used for sequencing analysis.

Three sets of primers (N1: 5'-CTSTACACSTCSGGSWSSACSGG-3'; C1: 5'-GTGCASACGTTSGTCTCSGTSGG-3', C2: 5'-AGSRCSGASGGSACSGARTA CCA-3', C3: 5'-AGSRCGCTSGGSACSGARTACCA-3') were used to amplify L-prolyl-AMP ligase gene fragments. Various pairs of primers (N1: 5'-TGG GAGMGSRCSTGCCT-3', N2: 5'-CAGGCSRTSTCSCACMGSMT-3', N3: 5'-CA GGCSRTSAGCCACMGSMT-3'; C1: 5'-ATSTCSSWSGTSCCSGAGAADAT-3', C2: 5'-ATSTCSSWSGTSCCGCGCTGAADAT-3', C3: 5'-ATSTCSKTSGTSCCSG AGAADAT-3', C4: 5'-ATSTCSKTSGTSCCGCTGAADAT-3') were used for the amplification of L-prolyl-PCP dehydrogenase gene fragments. PCR and cloning were carried out as described above.

Southern blot analysis of genomic DNA

Gene fragments homologous to *redH*, *redM* and *redW* have been amplified earlier by PCR from the roseophilin producer.⁵ DNA fragments used as probes were labeled using the AlkPhos Direct labeling kit (GE Healthcare, Piscataway, NJ, USA). *S. griseoviridis* genomic DNA was digested with *Bam*HI and run on a 0.7% agarose gel. The DNA was transferred to nylon membranes and fixed. After the membranes were incubated with probes at 65 °C for 16 h, they were washed and then developed using the ECF detection kit (GE Healthcare). Images were obtained with an FLA2000 image reader (Fujifilm, Tokyo, Japan).

Cloning of a prodigiosin biosynthesis gene cluster

A cosmid library was constructed from the roseophilin producer. Genomic DNA of *S. griseoviridis* 2464-S5 was partially digested with *Sau*3AI and ligated to the pWE15 cosmid vector (Stratagene, La Jolla, CA, USA). The ligated DNAs were packaged using Gigapack III XL Packaging Extract (Stratagene) and used to infect *E. coli* XL1-blue MRF'. *E. coli* colonies obtained by transfection with the cosmids were then screened for the presence of a *redW*-like gene by colony hybridization, using the PCR product labeled with the AlkPhos Direct labeling kit (GE Healthcare) as a probe. Selected positive clones were subsequently analyzed by Southern hybridization with the same probe, and cosmid 31 was unambiguously found to contain the *redW*-like sequence. A *Sac*I

6.0-kbp fragment of the cosmid was sequenced to identify genes homologous to *redD*, *redX*, *redW* and *redY*. To isolate the biosynthesis gene cluster, cosmids were selected using *redM-*, *redH-* and *redN-*like genes as probes, and DNA sequencing was carried out using a *SacI* 21.0-kbp fragment of cosmid16, a *SalI* 5.0-kbp fragment of cosmid16, a *SacI* 8.0-kbp fragment of cosmid10 and a *NcoI* 7.0-kbp fragment of cosmid11 (Figure 4).

Gene disruption in S. coelicolor

The disruption plasmid, pWHM3-DIS, was based on pWHM3, an *E. coli/ Streptomyces* shuttle vector (7.2 kbp) containing a thiostrepton-resistant gene.¹⁵ pWHM3 was digested with *Bcl*I and a 1.4-kbp fragment containing a part of the replication region in *Streptomyces* was removed. The remaining two *Bcl*I fragments (4.7 kbp and 1.1 kbp) were ligated to construct pWHM3-DIS.

The *redN* gene was disrupted by a double crossover event (Figure 5). A 2.7 kbp *XbaI/Hind*III fragment of *redN* contained both the upstream and downstream regions. The *redN* gene fragment was amplified by PCR with *S. coelicolor* genomic DNA and one set of primers with an additional restriction site (5'-TGCTCTAGATTCGACCTCGTCCACTACCTGCAG-3' and 5'-ACC <u>AAGCTTGTCGCCGGGGGTGAGGGGAGGGGAGGGCAG-3'</u>). The PCR product and *XbaI/Hind*III-digested pGEM-11Z were ligated and introduced into *E. coli*. A plasmid carrying the appropriate DNA fragment was selected. After sequence confirmation, the plasmid was digested with *BgI*II. A 0.7 kbp *BgI*II DNA fragment in the *redN* gene was removed and the remaining 5.0-kbp fragment containing pGEM-11Z was self-ligated. A 2.0 kbp *XbaI/Hind*III fragment obtained from the plasmid was subcloned into the same site of pWHM3-DIS to give pRedN-DIS.

The disruption plasmid, pRedN-DIS, was introduced into *S. coelicolor*, and thiostrepton-resistant colonies were selected. After the selected colonies were cultivated for 3 days, an aliquot of the culture was plated on ATCC medium #5 plates to obtain spores. Diluted spores were plated on ATCC medium #5 plates and thiostrepton-sensitive colonies were selected. Genomic DNA was prepared from the transformant and subjected to PCR analysis with the same primers that were used to amplify a 2.7 kbp DNA fragment of the *redN* gene.

Expression of *rphN* in the *redN* disruptant

To express the *rphN* gene in the *redN* disruptant, *S. coelicolor* $\Delta redN$, the *rphN*carrying plasmid, pWHM-rphN, was constructed. To obtain the entire *rphN* gene, PCR amplification was carried out under a standard condition with KOD-DNA polymerase (Toyobo, Osaka, Japan) using a pair of primers with an additional restriction site (5'-TGC<u>TCTAGA</u>ATGACCCACATCATGACC-GACCGT-3' and 5'-ACC<u>AAGCTTTCAGGCCGG</u>ATCGAGGGGGACGCC-3'). After sequence confirmation, the *XbaI*/*Hind*III DNA fragment was inserted into the same sites of pWHM860¹⁷ to give pWHM-rphN, in which the *rphN* gene was expressed under the control of the *ermE*^{*} promoter.

pWHM-rphN was introduced into S. coelicolor $\Delta redN$, and the transformants were cultivated at 27 °C for 6 days on a rotary shaker in 500-ml Erlenmeyer flasks containing 100 ml of production medium. The medium



Figure 7 Comparison of the prodigiosin biosynthesis gene clusters between S. coelicolor (red) and S. griseoviridis (rph).

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contained sucrose 103 g, glucose 10 g, yeast extract (Difco, Detroit, MI, USA) 5 g, Casamino acids (Difco) 0.1 g, sodium hydrogen L-glutamate monohydrate 3 g, K₂SO₄ 0.25 g, MgCl₂.6H₂O 10.12 g, NaNO₃ 0.06 g, TES (*N*-tris(hydroxymethyl)methyl-2-aminoethanesulfonic acid) buffer 5.73 g, 0.5 % KH₂PO₄ 10 ml, $5 \le CaCl_2 4$ ml, $1 \le NaOH 7$ ml and a trace element solution of 2 ml in 1000 ml distilled water (pH 7.2). The trace element solution consisted of ZnCl₂ 0.004%, FeCl₂.6H₂O 0.02%, CuCl₂.2H₂O 0.001%, MnCl₂.4H₂O 0.001%, Na₂B₄O₇.10H₂O 0.001% and (NH₄)₆Mo₇O₂₄.4H₂O 0.001%. The fermentation broth (100 ml) was centrifuged and mycelium was extracted with acetone. After evaporation, an aqueous concentrate was extracted with ethyl acetate. The extract was dissolved in methanol and the productivity of prodigiosins was examined by reversed-phase HPLC (Senshu Pak PEGASIL ODS 4.6 i.d.×250 mm, Senshu Scientific, Tokyo, Japan) with methanol–acetonitrile–trifluoroacetic acid–water (80:3:0.15:17) at a flow rate of 1.0 ml min⁻¹.

Nucleotide sequence and accession number

DNA sequencing was carried out with an automatic DNA sequencer (3100 Genetic Analyzer, Applied Biosystems, Foster City, CA, USA). Nucleotide sequences reported here have been deposited in the GenBank, DDBJ and EMBL databases under accession number AB469822 (*rph* gene cluster).

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ORIGINAL ARTICLE

Chitinase inhibitors: extraction of the active framework from natural argifin and use of *in situ* click chemistry

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In situ click chemistry is a target-guided synthesis technique for discovering potent protein ligands by assembling azides and alkynes into triazoles inside the affinity site of a target protein. We report the rapid discovery of a new and potent inhibitor of bacterial chitinases by the use of *in situ* click chemistry. We observed a target-templated formation of a potent triazole inhibitor of the chitinase-catalyzed chitin hydrolysis, through *in situ* click chemistry between a biologically active azide-containing scaffold and structurally unrelated alkyne fragments. Chitinase inhibitors have chemotherapeutic potential as fungicides, pesticides and antiasthmatics. Argifin, which has been isolated and characterized as a cyclopentapeptide natural product by our research group, shows strong inhibitory activity against chitinases. As a result of our efforts at developing a chitinase inhibitor from an azide-bearing argifin fragment and the application of the chitinase template and a library of alkynes, we rapidly obtained a very potent and new 1,5-disubstituted triazole inhibitor against *Serratia marcescens* chitinase (*Sm*Chi) B. The new inhibitor expressed 300-fold increase in the inhibitory activity against *Sm*ChiB compared with that of argifin. To the best of our knowledge, our finding of an enzyme-made 1,5-disubstituted triazole, using *in situ* click chemistry is the second example reported in the literature.

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Keywords: argifin; chitinase; in situ click chemistry; target-guided synthesis; triazole

INTRODUCTION

Chitin, the second most abundant polysaccharide in nature, is a constituent of fungal cell walls, the exoskeletons of crustaceans and insects, and the microfilarial sheaths of parasitic nematodes.¹⁻³ Accumulation of chitin by organisms is modulated by chitin synthasemediated biosynthesis and by chitinase-mediated hydrolytic degradation. Thus, chitinases are expected to be specific targets for antifungal, insecticidal and antiparasitic agents (for a review, see Andersen et al.⁴). They also offer significant potential for the treatment of asthma and other diseases in humans.⁵ Argifin (1) (Figure 1),⁶⁻⁸ a member of a family of natural cyclic peptides produced by microorganisms, was isolated in our laboratory from the cultured broth of the fungal strain (Gliocladium sp. FTD-0668), and was found to be a potent inhibitor of blowfly⁶ and Serratia marcescens chitinases (SmChi).⁹⁻¹² Recently, argifin complexes with fungal (Aspergillus fumigatus), human and bacterial chitinases have been resolved by X-ray crystallography.^{11,12} These studies revealed that there are at least four conserved hydrogenbond interactions between the N^{\odot} -methylcarbamoyl-L-arginine moiety and the polar groups arrayed in the hydrolytic pocket of the family 18 chitinases examined to date. The remarkable fidelity of the hydrogen-bonding network between the chitinases and the argifin ligand implicates its critical role in revealing the micro- to nanomolar range of inhibition. In fact, van Aalten and co-workers revealed through X-ray analysis that the ability of the N^{ω} -methylcarbamoyl group to penetrate fully into the active-site pocket of chitinases strongly correlated with the inhibition of chitin hydrolysis.¹³ Hence, we concluded that the N^{ω} -methylcarbamoyl-L-arginine core represents an ideal anchor to derivatize and elaborate better chitinase inhibitors.

Herein we report our early results of the design and simplification of azide-bearing N^{ω} -methylcarbamoyl-L-arginine substrate as a smaller analog of macrocyclic peptide natural product **1** and the use of target-guided synthesis (TGS) (for reports of TGS, see Rideout,¹⁴ Rideout,¹⁵ Ingelese and Benkovic,¹⁶ Boger *et al.*,¹⁷ Maly *et al.*,¹⁸ Nicolaou *et al.*,¹⁹ Greasley *et al.*,²⁰ Nguyen and Huc,²¹ Nicolaou *et al.*,²² Kehoe *et al.*,²³ Poulin-Kerstein and Dervan,²⁴ and Hu *et al.*²⁵) for the screening of new and more potent chitinase inhibitors, using the 1,3-dipolar cycloaddition²⁶ between an azide ligand and a library of acetylenes. The *in situ* click chemistry for drug discovery is

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Figure 1 Argifin (1) and the ${\rm IC}_{50}$ values against Serratia marcescens chitinases A, B and C_1.

dependent on irreversibly reacting reagents that are inert under physiological conditions,²⁷ as shown earlier by the discovery of highly potent inhibitors of acetylcholine esterase,^{28–31} carbonic anhydrase II³² and HIV-1 protease.³³

RESULTS AND DISCUSSION

Preparation of the azide-bearing inhibitors

Initially, we investigated the appropriate inhibitors to develop in situ click chemistry from the N^o-methylcarbamoyl-L-arginine scaffold. Although a simple arginine-derived inhibitor 2 discovered by van Aalten and co-workers had been reported independently,¹³ it showed low inhibitory activity against SmChi in contrast to argifin 1, which was examined by our group (Figure 2). We synthesized the azide-bearing inhibitor 3 as a reactive scaffold for capturing complementary acetylenic reagents to form triazole-linked inhibitors by TGS. Competition assay with 4-methylumberiferyl diacetylchitobiose $(4-MU-(GluNAc)_2)^{34,35}$ showed that this azide-bearing inhibitor 3 expressed a low inhibitory activity similar to that of the azide-lacking inhibitor 2, which are in striking contrast to the potency of the natural product 1. Hence, amide derivatives of azide 3 with amines other than methylamine were made and tested to see whether the binding could be restored to a level that would make azide 3 a sufficiently good anchor at the active site, to be used for the capture of alkyne-bearing candidates through in situ triazole formation. Fortunately, the dibenzylamide analog 5 of azide 3 emerged as a potent inhibitor (0.045 and 0.58 µM IC50 values against SmChiA and B, respectively). Interestingly, the IC₅₀ value against SmChiB of 5 was 10-fold stronger than that of parent 1. The monobenzylamide 4 was also active, but less than 5. As seen in Figure 2, compounds 2-5 can be ranked by inhibition constants as $5 > 4 > 3 \approx 2$. We therefore used potent azide analog 5 as a target 'anchor' molecule for in situ click chemistry.

Screening for enzyme-templated reaction

The *in situ* click chemistry experiments were carried out in parallel in 96-well microtiter plates to explore the chitinase-accelerated reaction using a mixture of *Sm*Chi A, B and C₁ (Sigma Co., Tokyo, Japan). Utilization of the mixed *Sm*Chi has the advantage of accelerating the identification of new inhibitors against each isozyme of chitinase through a one-off screening. Although a single isozyme or multiple isozymes of chitinase may participate in the formation of triazoles under this particular screening condition, the identification of the actual templating isozyme or isozymes can be determined in a follow-up assay using separate isozymes. Consequently, azide 5 (100 μ M) and 71 structurally diverse alkynes (300 μ M) were incubated in the presence



Figure 2 Structures and IC_{50} values of $\textit{N}^{\varpi}\text{-methylcarbamoyl-}\text{L-arginine-derived inhibitors.}$

of *Sm*Chi A, B and C₁ (9.6 mUnit ml⁻¹) in 10% methanol containing phosphate buffer solution at pH 7.0 (Scheme 1). The formation of the triazole products was monitored by HPLC and mass spectrometry by selected ion recording detection (LCMS-SIR, also known as HPLC and mass spectrometry in selected ion monitoring (LCMS-SIM)) after 20 h at 37 °C. After analysis of each reaction mixture, only alkyne **6** (IC₅₀ > 30 μ M) was sufficiently accelerated in its cycloaddition with azide **5** in the presence of the enzymes to yield a detectable amount of triazole **7** (at this point not distinguished as to whether a *syn-* or *anti*-substituted triazole) in the background with great reproducibility by LCMS-SIR measurement (Figure 3). In effect, the chitinases had performed as a reaction vessel of molecular scale to create its own better inhibitor.

Preparations and inhibitory activities of triazole 7

Subsequently, azide 5 and alkyne 6 were subjected to copper(I)catalyzed azide-alkyne cycloaddition conditions (CuAAC)³⁶⁻³⁸ along with the ruthenium-catalyzed azide-alkyne cycloaddition reaction conditions (RuAAC),^{39,40} to prepare pure regioisomers of 7 allowing the identification of the regiochemistry of the triazole formed by the enzymes. As expected, pure 1,4- and 1,5-disubstituted triazole products (anti-7 and syn-7) were obtained (Scheme 2). Having both pure triazoles in hand, we turned our attention to the identification of the generated triazole analog by TGS and the participating isozymes of the enzyme for this *in situ* click chemistry. For the determination of IC₅₀ values against each SmChi isozyme, anti- and syn-7 were assessed in a competition assay with 4-MU-(GluNAc)₂. As shown in Figure 4, the inhibitory activities of both regioisomers of 7 against SmChi A and C1 were almost the same as those of 5. On the other hand, syn-7 displayed high inhibitory activity against SmChi B (IC50 value of 0.022 µM), which is approximately 30-fold stronger than that of 5 (approximately 300-fold improved potency as compared with the natural product 1). These results strongly indicate that syn-7 is most likely formed in situ by the SmChi B isozyme in the enzyme mixture.



Scheme 1 SmChi templated in situ click chemistry protocol and the guided triazole analog.



Figure 3 Results of *in situ* click chemistry between **5** and **6**, monitored by LCMS-SIR. (a) Authentic sample of **7** from thermal reaction (100 °C, 12 h), apparently single peak (4.9 min) of **7** (*anti: syn*=3: 2) was observed; (b) reaction (37 °C, 20 h) between **5** (100 μ M) and **6** (300 μ M) in the presence of *Sm*Chi (9.6 mUnit ml⁻¹); (c) without *Sm*Chi (background reaction).

Acceleration effect of *syn*-triazole formation through *Sm*ChiB

Analysis of *syn-anti* selection for the *in situ* screening by LCMS-SIR revealed that a combination of azide **5** and alkyne **6** had led to the accelerated formation of *syn-7* in the presence of pure $(His)_6$ -*Sm*ChiB in an enzyme-dose-dependent manner (Figure 5). Moreover, no *syn*-triazole formation was observed in the control incubation containing *Sm*Chi B and the same azide and alkyne in the presence of a natural product argadin acting as a potent inhibitor (see Supplementary Information for structure)⁴¹ (IC₅₀ values against *Sm*ChiB, $33 \pm 2.8 \text{ nM}$), thereby validating *syn-7* as an *in situ* hit and confirming that its formation required the enzyme active site to be accessible. Interestingly, the regioisomer *anti-7*, which is predominantly formed



Scheme 2 Preparation of *anti* and *syn-7*. TBTA=*tris*-(benzyltriazolylmethyl) amine, THF=tetrahydrofuran.



Figure 4 IC₅₀ values of anti and syn-7 against SmChiA, B and C1.

under pure thermal Huisgen conditions, is less active against *Sm*ChiB than the 'anchor' molecule **5**, which probably presents the $-CH_2-N_3$ group in a unique position when **5** and the protein form their complex. At this point, the chitinase–**5** complex is most likely a single entity presenting the azide properly to the 'well-suited' alkyne ligand and correctly binding to the complex so that a *syn*-triazole selectively clicks into its existence. Now the two small molecules have become one, and this enzyme-templated molecule invariably binds more strongly than either component alone. The high kinetic barrier would effectively lower **5** and **6** until they are attracted to each other to form the thermodynamically unfavorable *syn*-isomer.



Figure 5 Identification of *syn-anti* selection for *in situ* click chemistry between **5** (100 μ M) and **6** (300 μ M), monitored by LCMS-SIR under different conditions from Figure 3. (a) Authentic sample of **7** from thermal reaction (100 °C, 12 h): *syn-***7** (9.0 min) and *anti-***7** (9.5 min) (2:3 ratio); (b) reaction with pure (His)₆-*Sm*ChiB (192 μ g ml⁻¹; 37 °C, 20 h); (c) (His)₆-*Sm*ChiB (96 μ g ml⁻¹; 37 °C, 20 h); (d) (His)₆-*Sm*ChiB (48 μ g ml⁻¹; 37 °C, 20 h); (e) (His)₆-*Sm*ChiB (192 μ g ml⁻¹), argadin (100 μ M; 37 °C, 20 h); and (f) without enzyme (background reaction).

Conclusion and remarks

We discovered a highly active inhibitor of chitinase through *in situ* click chemistry. The strategy used an azide substituent appended to an active domain excised, as it were, from the more complex natural macrocyclic peptide **1**. The *Sm*Chi, which in this case was specifically *Sm*ChiB, acts as both mold and template for triazole formation between a unique pair of azide and alkyne fragments. Indeed, a number of analogs, based on bioactive molecules, would have to be synthesized to reveal the SAR for the affinity of a specific target

molecule (e.g. enzymes) and to identify superior materials for traditional lead discovery. In the process of *in situ* click chemistry, the highly exergonic nature of triazole formation makes the process completely irreversible and thereby locks in unique information, a kind of embedded message of the encounter. More practically, it allowed us to discover a lead template for the discovery of a selective chitinase inhibitor directed toward the functions of *Sm*Chi, without the need for long and costly analog syntheses. Ongoing studies are focused on optimizing the structure and further evaluating

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bioactivities against various types of chitinases, as well as promoting efforts to seek entirely new inhibitors, not only for chitinases but also for thorough application and refinement of the basic *in situ* methodology.

METHODS

Details of the experimental procedures and characterization of the new compounds, list of the acetylenes library, the expression of each *Serratia marcescens* chitinases and IC_{50} determination can be found in the Supplementary Information.

In situ click chemistry experiment using a mixture of SmChiA, B and C₁

Azide 5 (5µl; 2.0 mM in MeOH) was diluted in 0.1 M phosphate buffer (pH 7.0) (60 µl). Subsequently, the required alkynes (5 µl; 6.0 mM in MeOH) were added to the azide solution, followed immediately by the addition of a solution of SmChi A, B and C1, which was purchased from Sigma Co. (C-7809-1UN) (for the ratio of the SmChi isozymes, see Supplementary Figure 10 in the Supplementary Information), in 0.1 M phosphate buffer $(30 \,\mu\text{l}; 32 \,\text{mUnit}\,\text{ml}^{-1})$ to give final concentrations of 9.6 mUnit ml⁻¹ SmChi, $100\,\mu\text{M}$ azide 5 and 300 μM of each alkyne (Supplementary Figure 7) in 10% MeOH/phosphate buffer (100 µl). After mixing thoroughly, the reaction mixtures were incubated at 37 $^\circ C$ for 20 h, then diluted with 100 μl MeOH to restrain the enzyme activity and injected directly into the LC/UV-MS instrument to perform LCMS-SIR analysis (Column: Senshu Pak Pegasil (Senshu Scientific Co., Tokyo, Japan) ODS 2ø×50 mm; conditions for HPLC: gradient 10% MeCN (0.05% TFA)/H₂O (0.1% TFA) to 100% MeCN (0.05% TFA) over 8 min, flow 0.3 ml min⁻¹, detect 200–400 nm, temp 20 °C; MS-SIR: cone voltage 60 V, source temp 110 °C, desolvation temp 350 °C, selected mass 704 [M+H]⁺). The triazole products were identified by their retention times and molecular weights. Control experiments without chitinases (azide 5 (5 µl; 2.0 mM in MeOH) and alkynes (5 µl; 6.0 mM in MeOH) in 0.1 M phosphate buffer (pH 7.0) (90 µl) at 37 °C for 20 h) were run consecutively.

One unit of chitinases can liberate 1.0 mg of *N*-acetyl-D-glucosamine from chitin per hour at pH 6.0 at in a 2-h assay.

Expression and purification of (His)₆-SmChiB

SmChiB gene was amplified by PCR and cloned in frame to pTrcHis B vector (Invitrogen, Tokyo, Japan) to give pTrcHisSmChiB. E. coli strain DH5a harboring plasmid pTrcHisSmChiB was harvested and disrupted by the same procedure used for expression of crude SmChiB. After centrifugation, the supernatant was applied to HisTrap Chelating HP column (Amersham Biosciences, Tokyo, Japan), and the bound proteins were eluted by a stepwise gradient of imidazole in 100 mM phosphate buffer. (His)₆-SmChiB was eluted at a concentration of over 150 mM imidazole and showed a molecular mass of 53 kDa on SDS-PAGE (Supplementary Figure 10). The combined fraction was concentrated and imidazole was removed using HiTrap Desalting column (Amersham BioSciences). The protein concentration (320 µg ml⁻¹) was determined by Bradford method using protein quantification kit-rapid (Dojindo, Kumamoto, Japan) with BSA as a standard protein (0.80 absorbance at 595 nm for diluted protein solution (×10)) (Supplementary Figure 10). The chitinolytic activity of purified (His)₆-SmChiB was assayed by the procedure in the Supplementary Information (Supplementary Figure 10).

In situ formation of syn-7 by SmChiB template reaction

A chromatographic and analytical method different from *in situ* click screening was used for determining the *syn-anti* selection of the *in situ* click chemistry product 7 by (His)₆-SmChiB. The assignment was accomplished by comparing the retention times of the *in situ* products with authentic samples prepared by the thermal reaction of triazole formation.

Azide **5** (5 µl; 2.0 mM in MeOH) was diluted in 0.1 M phosphate buffer (pH 7.0) (60 µl). Subsequently, the alkyne **6** (5 µl; 6.0 mM in MeOH) was added to the azide solution, followed immediately by the additon of a solution of (His)₆-*Sm*ChiB (320 µg ml⁻¹) in 0.1 M phosphate buffer (30 µl) to give a final concentration of 96 µg ml⁻¹(His)₆-*Sm*ChiB, 100 µM azide **5** and 300 µM alkyne

6 in 10% MeOH/phosphate buffer (100 µl). After mixing thoroughly, the reaction mixtures were kept at 37 $^{\circ}\mathrm{C}$ for 20 h, then cooled to 20 $^{\circ}\mathrm{C}$ and diluted with 100 µl MeCN to restrain the enzyme activity and injected directly into the LC/UV-MS instrument to perform LCMS-SIR analysis (Column: Develosil (Nomura Chemical Co., Aichi, Japan) C30-UG-5, 2ø×150 mm; conditions for HPLC: isocratic 29% MeCN(0.05% TFA)/H₂O(0.1%TFA), flow 0.3 ml min⁻¹, detect 200-400 nm, temp 20 °C; MS-SIR: cone voltage 95 V, source temp 110 °C, desolvation temp 350 °C, selected mass 704 [M+H]⁺). The triazole products were identified by their retention times and molecular weights. In concurrence, experiments to investigate the dose dependency of the enzyme for production of syn-7 were carried out under the same condition by using different concentrations of (His)₆-SmChiB (final concentrations 192 and $48 \,\mu g \,ml^{-1}$ (His)₆-SmChiB, respectively, 100 μ M azide 5 and 300 μ M of alkyne 6 in 10% MeOH/phosphate buffer). From these comparisons, dose-dependent production of syn-7 was clearly observed (Figure 5). LCMS-SIR peaks of synand anti-7 were confirmed by co-injection of authentic sample 7 (from thermal reaction) and a crude mixture of enzyme template reaction ((His)₆-SmChiB 192 µg ml⁻¹, 100 µм azide and 300 µм alkyne at 37 °C, 20 h) (Supplementary Figure 11). Control experiments in the presence of the known SmChiB inhibitor Argadin ^{41} (IC_{50}~33 \pm 2.8 \, \text{nm}) (final concentration: 192 $\mu g \, m l^{-1}$ SmChiB, 100 µM argadin, 100 µM azide 5 and 300 µM alkyne 6) and in the absence of enzyme (100 $\mu \rm M$ azide 5 and 300 $\mu \rm M$ alkyne 6) were run consecutively.

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Supplementary Information accompanies the paper on The Journal of Antibiotics website (http://www.nature.com/ja)

NOTE

JBIR-17, a novel trichostatin analog from *Streptomyces* sp. 26634

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Histone deacetylases (HDACs) play an important role in the epigenetic regulation of gene expression by catalyzing the removal of acetyl groups from lysine residue of histone protein, stimulating chromatin condensation and promoting transcriptional repression.^{1,2} HDACs are divided into four classes on the basis of their homology to yeast HDACs: class I (HDAC1, 2, 3 and 8), class IIa (HDAC4, 5, 7 and 9), class IIb (HDAC6 and 10), class III (SIRT1, 2, 3, 4, 5, 6 and 7) and class IV (HDAC11). As aberrant epigenetic changes are a hallmark of cancer, HDACs are a promising target for an anticancer drug. The inhibitors of HDACs can induce cell-cycle arrest, promote differentiation and stimulate tumor cell death. In fact, several HDAC inhibitors are currently in clinical trials both for solid and hematological malignancies.^{1,2} Therefore, we attempted to search new HDAC inhibitors. As a result, we isolated a novel compound designated as JBIR-17 (1) from Streptomyces sp. 26634 (Figure 1). We report herein the isolation, structure elucidation and biological activity of 1.

Streptomyces sp. 26634 was isolated from a leaf of Kerria japonica collected in Iwata, Shizuoka Prefecture, Japan, and cultured on a rotary shaker (220 r.p.m.) at 28°C for 4 days in a 500-ml Erlenmeyer flask containing 60 ml of a medium consisting of 4% β -cyclodextrin, 0.5% glycerol, 2% Pharmamedia (Traders Protein, Lubbock, TX, USA), 0.0005% CuSO₄·5H₂O, 0.0005% MnCl₂·4H₂O and 0.0005% ZnSO₄·7H₂O.

n-BuOH (37.5 ml) was added to the fermentation broth (60 ml) and shaken for 15 min. After centrifugation, the organic layer was evaporated *in vacuo*. The dried residue (107 mg) was subjected to reversed-phase medium-pressure liquid chromatography (Purif-Pack ODS 100, Moritex, Tokyo, Japan) and eluted with a MeOH–H₂O (5-100% MeOH) linear gradient system. The 70–90% MeOH eluate (3.7 mg) was further purified by reversed-phase HPLC using an

XBridge Prep C₁₈ column (5 μ m optimum bed density (OBD), 4.6 i.d.×250 mm, Waters, Milford, MA, USA) with 35% aqueous CH₃CN containing 0.2% formic acid (flow rate, 1 ml min⁻¹) to yield JBIR-17 (1, 0.9 mg; retention time (Rt) 9.7 min) and trichostatin A (2, 0.7 mg; Rt 10.3 min).³

The physicochemical properties of 1 are summarized in Table 1. Compound 1 was obtained as a colorless amorphous solid, and its molecular formula was determined to be $C_{20}H_{26}N_2O_5$ by HR-electrospray ionization (ESI)-MS. The IR spectrum revealed the characteristic absorptions of the aroyl and/or amide carbonyl (v_{max} 1652 cm⁻¹) and amide N-H (v_{max} 1597 cm⁻¹) groups. The structure of 1 was mainly determined by NMR spectral analyses as follows.

The direct connectivity between each proton and carbon was established by the heteronuclear single quantum coherence spectrum, and the ¹³C and ¹H NMR spectral data for 1 are shown in Table 2. A total of 20 signals were observed in the ¹³C NMR spectrum, consistent with the HR-MS data. These signals included three carbonyl (C-1, C-7, C-1'), four olefinic (C-2 to C-5) and six aromatic (C-8 to C-13) carbons. The proton spin couplings between two olefinic protons 2-H $(\delta_{\rm H}$ 5.91) and 3-H ($\delta_{\rm H}$ 7.28), between an olefinic proton 5-H ($\delta_{\rm H}$ 6.04) and a methyl proton 6-Me (δ_H 1.28) through a methine proton 6-H ($\delta_{\rm H}$ 4.36), and between two equivalent aromatic protons 9,13-H $(\delta_{\rm H} 7.84)$ and 10,12-H $(\delta_{\rm H} 6.69)$ on a *p*-disubstituted benzene ring were observed in a double-quantum filtered (DQF)-COSY spectrum as shown in Figure 2 (bold line). The constant time heteronuclear multibond correlation (CT-HMBC) experiment revealed the presence of $^1\text{H-}{^{13}\text{C}}$ long-range couplings from an N,N-dimethyl proton (δ_{H} 3.05) to an aromatic carbon C-11 (δ_C 155.2), from 9,13-H to C-11 and a carbonyl carbon C-7 ($\delta_{\rm C}$ 198.7), from 10,12-H to an aromatic carbon C-8 (δ_{C} 127.0), from 6-Me, 6-H and 5-H to C-7, from the

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Figure 1 Structures of JBIR-17 (1) and trichostatin A (2).

Table 1 Physicochemical properties of 1

Appearance	Colorless amorphous solid
Optical rotation ^a	[α] $^{25}_{ m D}{-}18.0^{\circ}$ (c 0.1, MeOH)
HR-ESI-MS ^b (m/z)	
Found	375.1880 (M+H)+
Calculated	375.1920 (C ₂₀ H ₂₇ N ₂ O ₅)
UV ^c λ _{max} nm (ε)	264 (24700), 336 (23500)
IR ^d (KBr) v _{max} cm ⁻¹	3415, 1653, 1597

Abbreviation: ESI, electrospray ionization.

^aOptical rotation was measured on a SEPA-300 polarimeter (Horiba, Kyoto, Japan).

^bHR-ESI-MS measurement was carried out on a LCT Premier XE mass spectrometer (Waters). ^cUV spectrum was measured on a COULTER DU730 UV/Vis Spectrophotometer (Beckman, Fullerton, CA, USA).

^dIR spectra was obtained on a FT-720 Fourier transform infrared spectrometer (Horiba).

Table 2 ¹³ C and ¹		data for	r 1
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Position	$\delta_{\mathcal{C}}$	δ_H
1	168.6	
2	117.5	5.91 (d, 15.4)
3	147.5	7.28 (d, 15.4)
4	132.5	
5	141.9	6.04 (d, 9.5)
6	40.9	4.36 (dq, 9.5, 6.9)
7	198.7	
8	127.0	
9, 13	130.7	7.84 (d, 8.8)
10, 12	111.2	6.69 (d, 8.8)
11	155.2	
1′	172.0	
2′	54.8	4.60 (br s)
3′	62.1	4.18 (br d, 10.5); 3.85 (br d, 10.5)
4-methyl	12.6	1.89 (s)
6-methyl	17.8	1.28 (d, 6.9)
11N, N-dimethyl	40.3	3.05 (s)

 ^{13}C (125 MHz) and ^1H (500 MHz) NMR spectra were taken on a NMR System 500 NB CL (Varian, Palo Alto, CA, USA) in CDCl₃, and the solvent peak was used as an internal standard (δ_C 77.0, δ_H 7.26).

vinyl methyl proton 4-Me (δ_H 1.89) to three olefinic carbons C-3 (δ_C 147.5), C-4 (δ_C 132.5) and C-5 (δ_C 141.9), and from 2-H and 3-H to an amide carbonyl carbon C-1 (δ_C 168.6). The stereochemistries of



Figure 2 Key correlations in DQF-COSY (bold line) and CT-HMBC (arrow) spectra of $1. \label{eq:correlation}$

two olefins were determined as 2*E* and 4*E* according to the coupling constant ($J_{2,3}$ =15.4 Hz) and the high-field-shifted ¹³C chemical shift at 4-Me ($\delta_{\rm C}$ 12.6). Thus, the partial structure was elucidated as a trichostatic acid (**3**) moiety, and their ¹³C and ¹H NMR signals are superimposable with those of **2**³ and **3**⁴.

Additional substructure was elucidated as follows. A proton spin coupling between an α -methine proton 2'-H (δ_H 4.60; δ_C 54.8) and oxymethylene protons 3'-H (δ_H 4.18, 3.85; δ_C 62.1) was observed. A long-range coupling from 3'-H to a carboxylic carbonyl carbon C-1' deduced that the remaining structure was a serine moiety, and the serine was assumed to attach to C-1 of trichostatin through an amide bond.

The linkage position and the absolute configuration of the serine moiety of 1 were confirmed as follows. To determine the absolute configuration of the serine moiety, Marfey's method was adopted. Compound 1 (0.8 mg) was hydrolyzed with 6 N HCl (0.2 ml) at 120°C overnight to obtain the serine residue. After acid hydrolysis, the reaction solution was adjusted to neutral pH and evaporated in vacuo. The residue was dissolved in an aqueous solution of 0.1 M NaHCO₃ (0.6 ml), and 10 mM N^{\alpha}-(5-fluoro-2,4-dinitrophenyl)-Lalaninamide (FDAA) in Me₂CO (0.6 ml) was successively added. The mixture was kept at 70°C for 10 min with frequent shaking. After work-up with the addition of 0.2 N HCl, the filtered reaction mixture was subjected to ultra performance liquid chromatography (UPLC) analysis (Acquity UPLC BEH C_{18} 1.7 µm, 2.1×50 mm, Waters; 10% aqueous CH₃CN containing 0.1% formic acid; flow rate, 0.3 ml min⁻¹). The authentic D- and L-serine were reacted with FDAA in the same manner as described above. The serine residue obtained from the hydrolysate was determined to be L-serine (Rt 8.6 min; L-Ser, 8.5 min; D-Ser, 9.8 min). To confirm the linkage position of serine moiety, 1 was semi-synthesized from 2 as shown in Figure 3. Briefly, 2 (7.5 mg) was converted to 3 (3.4 mg) by HClO₄. Compound 3 was coupled with an O-t-butyl-L-serine t-butyl ester in the presence of PyBOP and N,N-diisopropylethylamine followed by deprotection in acidic condition to yield an L-serine adduct of 3 (0.9 mg). This synthetic compound showed an identical ¹H NMR spectrum to that of naturally isolated 1 from Streptomyces sp. 26634.

To evaluate inhibitory activity for HDACs of 1, we used the reporter gene assay system using a luciferase gene as described earlier.^{5,6} The human embryonic kidney 293T cells, transformed with the luciferase reporter gene driven by the cytomegalovirus promoter, produced 2.5 times more luciferase compared with the untreated control, when they were treated with 1 at a concentration of 30 μ M. Furthermore, to clarify the selectivity against HDAC subtypes, 1 was tested in the HDAC inhibitory activity using HDAC1 (class I), 4 (class IIa) and 6 (class IIb) enzymes of 293T cell origin, which are usually used as the representative HDACs among each HDAC subtype.⁷ Compound 1 showed inhibitory activity against HDAC4 and 6 with IC₅₀ values of 69 and 4.7 μ M, respectively, but no activity against HDAC1 at a

Figure 3 Scheme of chemical conversion from **2** to **1**. (a) 1.5 N HClO₄ aq, 50° C, overnight. (b) *O*-*t*-butyl-L-serine *t*-butyl ester, PyBOP, *N*,*N*-diisopropulethylamine in CH₂Cl₂/DMF (*N*,*N*-dimethylformamide), room temperature, 2 h. (c) 90% aqueous trifluoroacetic acid, room temperature, 1.5 h.

concentration of 100 μ M. In contrast, **2** showed strong, but not selective, inhibitory effects against these HDACs (IC₅₀ values, 18, 30 and 92 nM against HDAC1, 4 and 6, respectively) as reported earlier.^{8,9} These results indicated that **1** selectively inhibited HDAC6 compared with HDAC1 and 4. HDAC6 is a cytoplasmic enzyme that regulates many important biological processes, including cell migration, immune synapse formation, viral infection and the degradation of

misfolded proteins. Furthermore, HDAC6 deacetylates tubulin, Hsp90 and cortactin.^{10–12} The diverse functions of HDAC6 suggest that it is a potential therapeutic target for a wide range of diseases. Thus, JBIR-17 could be a valuable tool for the studies of HDAC6 and enzymatic property among HDAC subtypes.

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OBITUARY

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Hans Zähner (7 June 1929–18 December 2008)

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 \mathbf{P} rofessor Hans Zähner passed away on 18 December 2008 at the age of 79 after a prolonged illness.

Hans Zähner was born on 7 June 1929 in Zürich, Switzerland. He studied agricultural science at the Eidgenössische Technische Hochschule (ETH) in Zürich from 1949 to 1953 and worked with Professor Ernst Gäumann at the Institut für Spezielle Botanik at ETH, finishing his doctoral thesis on phytopathology in 1954. From 1954 to 1958, he collaborated with Professor Leopold Ettlinger and Roger Corbaz at the same institute. Their aim was to screen microorganisms for new biologically active compounds and extend the screening from fungi to include actinomycete strains. This period marked the beginning of a very successful collaboration with Professor Vlado Prelog and Walter Keller-Schierlein from the Institute of Organic Chemistry at ETH in the isolation and structure elucidation of new compounds. About 20 000 actinomycete strains were tested to see if they produced antibiotics. Besides the identification of 1100 producers of known antibiotics, 20 new compounds were isolated. The structures of 10 were determined and published in the series 'Stoffwechselprodukte von Actinomyceten': narbomycin, angolamycin, nonactin, foromacidins, echinomycin, granaticin, megacidin, acetomycin, cinerubins and holomycin.

In 1958, Hans Zähner took responsibility for microbiological research in the group led by Professor Ettlinger, who was appointed to the professorship of bacteriology at ETH. Together with Ralph Hütter and Elisabeth Bachmann at the Institut für Spezielle Botanik and his colleagues Professor Prelog and Walter Keller-Schierlein from the Institute of Organic Chemistry, he intensified the search for new antibiotics. Although the number of strains tested in screening programs was reduced, new methods for isolating actinomycetes from soil samples were developed, new test organisms and cultivation methods for these strains were used, and new screening methods were devised. These methodological changes led to a change in the nature of the detected compounds and resulted in the isolation of numerous new substances; for example, actinomycin Z, actiphenol, acumycin, aranciamycin, boromycin, ferrichrysin, granaticin B, lankamycin, lankacidin, manumycin, naphthocyclinones, naphthomycin, scopamycins and venturicidin B. Hans Zähner's scientific success led to his habilitation in 1960.

During investigation into the optimization of ferrimycin production, it was observed that the antibiotic activity of ferrimycin could be reversed by cultures of other actinomycete strains. This antagonism was specific to ferrimycin; other antibiotics like streptomycin and the tetracyclines did not share this property, and this was the beginning of Hans Zähner's successful studies of iron-chelating metabolites—the siderophores—resulting in the industrial production of desferri-



ferrioxamine B, which was introduced on the market as Desferal in 1962 by Ciba (Basel, Switzerland).

In 1964, he moved from ETH Zürich to the University of Tübingen in Germany to a newly created Chair in microbiology. There he founded the Institute of Microbiology and continued his work on screening for novel secondary metabolites. The screening for ironchelating compounds, for antibiotics that act exclusively on chemically defined media and for antibiotics that cause morphological changes in fungi, was continued successfully and was completed by a chemical TLC-based screening program. Further remarkable compounds were detected by Hans Zähner and his group, including, among others, avilamycins, bafilomycins, the nikkomycins, phosphinothricin, tetracenomycins and urdamycins. At the beginning of his career, he started the communication series 'Microbial Products of Microorganisms', which, in July 1994, the year of his retirement, reached Part No. 269. Under his supervision, 145 students obtained their doctoral degree in the Institute of Microbiology at the University of Tübingen.

Hans Zähner was a member of the Editorial Board of *The Journal* of *Antibiotics* from 1972 until 1995 and will sorely be missed by his friends and colleagues in the microbiological community.

He married his wife, Hedi, in 1954 and they have four children and three grandchildren. We all wish to express our deepest sympathy to Hedi and her family.

Hans-Peter Fiedler Institute of Microbiology, University of Tübingen, Germany